Purification, Biochemical Characterization, and Implications of an Alkali-Tolerant Catalase from the Spacecraft-Associated and Oxidation-Resistant *Acinetobacter gyllenbergii* 2P01AA

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Abstract

Herein, we report on the purification, characterization, and sequencing of catalase from Acinetobacter gyllenbergii 2P01AA, an extremely oxidation-resistant bacterium that was isolated from the Mars Phoenix spacecraft assembly facility. The Acinetobacter are dominant members of the microbial communities that inhabit spacecraft assembly facilities and consequently may serve as forward contaminants that could impact the integrity of future life-detection missions. Catalase was purified by using a 3-step chromatographic procedure, where mass spectrometry provided respective subunit and intact masses of 57.8 and 234.6 kDa, which were consistent with a small-subunit tetrameric catalase. Kinetics revealed an extreme pH stability with no loss in activity between pH 5 and 11.5 and provided respective k_{cat}/K_m and k_{cat} values of $\sim 10^7 \text{ s}^{-1} M^{-1}$ and 10^6 s^{-1} , which are among the highest reported for bacterial catalases. The amino acid sequence was deduced by in-depth peptide mapping, and structural homology suggested that the catalases from differing strains of A. gyllenbergii differ only at residues near the subunit interfaces, which may impact catalytic stability. Together, the kinetic, alkali-tolerant, and halotolerant properties of the catalase from A. gyllenbergii 2P01AA are significant, as they are consistent with molecular adaptations toward the alkaline, low-humidity, and potentially oxidizing conditions of spacecraft assembly facilities. Therefore, these results support the hypothesis that the selective pressures of the assembly facilities impact the microbial communities at the molecular level, which may have broad implications for future life-detection missions. Key Words: Catalase-Kinetics-Sequence-Stability-Acinetobacter-Spacecraft—Planetary protection. Astrobiology 15, 291–300.

1. Introduction

THE Acinetobacter are potential microbial forward contaminants (Derecho *et al.*, 2014) that may impact the integrity of future life-detection missions on Mars (Space Studies Board, 2006). Molecular and cultivation studies show that the Acinetobacter are among the dominant members of the microbial communities that inhabit spacecraft and the associated assembly facilities (La Duc *et al.*, 2003, 2012; Ghosh *et al.*, 2010; Vaishampayan *et al.*, 2010). Hence, these microorganisms carry a high relative potential to be transported along with assembled spacecraft and serve as sources of false-positive signals of life. Crucially, molecular community analyses conducted during the Mars Phoenix lander assembly (Ghosh *et al.*, 2010; Vaishampayan *et al.*, 2010) indicated that the Acinetobacter possess tolerances toward (and potentially proliferate under) the assembly conditions despite the strict cleaning regimes, which included surface wiping with isopropanol and floor cleansing with alkaline detergents (Frick *et al.*, 2014). Specifically, these studies show that the operational taxonomic units (OTUs) associated with the *Acinetobacter* increased by ~9-fold throughout the assembly process, whereas, in contrast, the overall OTUs that represent the total bioburden decreased by ~3-fold (Vaishampayan *et al.*, 2010). Hence, these combined studies suggest that non-spore-forming bacteria such as the *Acinetobacter* may pose cleanliness issues for future life-detection missions.

The Acinetobacter are broadly associated with soil, water, and clinical environments (Gerischer, 2008) and have now been isolated and genetically detected in multiple spacecraft-associated environments, including the assembly facilities for space telescopes (Moissl-Eichinger *et al.*, 2013), the

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assembly facilities for Mars-destined spacecraft (La Duc et al., 2012), on the surface of the preflight Mars Odyssey orbiter (La Duc et al., 2003, 2004b), and on surfaces and in the drinking water of the International Space Station (Castro et al., 2004; La Duc et al., 2004a). Recently, we reported on the characterization of Acinetobacter gyllenbergii 2P01AA, which was isolated from the floor of the Mars Phoenix assembly facility. Our studies show that A. gyllenbergii 2P01AA displays an extremotolerance toward hydrogen peroxide (H_2O_2) that is perhaps the highest known among Gram-negative and non-spore-forming bacteria (no loss in

 $100 \text{ m}M \text{ H}_2\text{O}_2$) (McCoy *et al.*, 2012; Derecho *et al.*, 2014). Biochemical analyses showed that the extremotolerance is related to the enzymatic degradation of hydro- and alkylperoxides, as well as proteomic factors associated with energy management, protein synthesis and folding, membrane transport, and nucleotide metabolism (Derecho *et al.*, 2014). Accordingly, the role of catalase, which is an enzyme that

degrades H_2O_2 , has been shown to be a key factor in the extremotolerance of spacecraft-associated microorganisms. For example, studies on A. gyllenbergii 2P01AA reveal very high catalase specific activities $(1800 \pm 350 \text{ Units/mg})$ that are ~130% of that of *Deinococcus radiodurans* R1 and ~50% of that of Vibrio rumoiensis S-1^T (Derecho et al., 2014), which are radiation-resistant bacteria (Daly, 2009) and oxidation-resistant bacteria (Yumoto et al., 1999), respectively. In fact, relatively high catalase specific activities have been found in all tested spacecraft-associated Acinetobacter, including all floor isolates from the Mars Phoenix assembly facility and from A. radioresistens 50v1, which was isolated from the surface of the preflight Mars Odyssey spacecraft (McCoy et al., 2012; Derecho et al., 2014). Studies on A. radioresistens 50v1 reveal that the catalase specific activities are inducible and increase by $\sim 50\%$ upon exposure to 1 mM H₂O₂ (McCoy et al., 2012). Further, catalases have been implicated in spore protection for the spacecraft-associated Bacillus pumilus SAFR-032 (Checinska et al., 2012). Thus, when considered together, these studies suggest that catalase is an important biochemical determinant for survival in spacecraft assembly facilities.

In this study, therefore, we describe the purification, characterization, and sequencing of a catalase from A. gyllenbergii 2P01AA and examine the implications toward survival in the assembly facilities. To date, there remain very few reports that focus on the biochemical characterizations of spacecraft-associated microorganisms (Gioia et al., 2007; Checinska et al., 2012; McCoy et al., 2012; Derecho et al., 2014), which is significant, as these types of analyses will assist in the interpretation of life-based detection experiments, such as those planned for the upcoming NASA Mars 2020 and ESA ExoMars 2018 missions. Hence, our current study represents the first reported purification and analysis of an enzyme from a spacecraft-associated microorganism. In summary, our results show that the catalase from A. gyllenbergii 2P01AA is extremely pH stable, alkali-tolerant, halotolerant, and possesses k_{cat}/K_m and k_{cat} values that are among the highest reported for bacterial catalases. Together, these properties are consistent with adaptations toward the alkaline, low-humidity, and potentially oxidizing conditions of spacecraft assembly facilities, and hence support the hypothesis that the assembly conditions impact the microorganisms at the molecular level.

2. Materials and Methods

2.1. Materials

Materials included Bugbuster Master Mix (EMD4Biosciences), Halt* Protease Inhibitor Cocktail (Thermo Scientific), bovine liver catalase (Calbiochem), sodium pyrophosphate (Sigma-Aldrich), HEPES (4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid) (VWR), NaCl (VWR), phosphate-buffered saline (PBS; $10 \times$ PBS: 100 mM potassium phosphate, 100 mM NaCl, pH 7.4) (VWR), and 1 cm disposable BRAND semi-micro UV-cuvettes (VWR). Nonstabilized 30% w/w hydrogen peroxide (Sigma-Aldrich) was utilized in the kinetic reactions to eliminate the impacts of the stabilizers (e.g., phenol, acetanilide, and sodium stanate): hydrogen peroxide solutions were stored as aliquots at -20° C. Lysogeny broth (LB) was prepared (per liter) by using 5.0 g of yeast extract (Becton Dickinson), 10.0 g of tryptone (Becton Dickinson), and 10.0 g of NaCl. All media were autoclaved at 121°C for 30 min, buffers and solutions were sterile filtered (0.22 μ m), and pure water (18 M Ω cm⁻¹) was used throughout.

2.2. Catalase purification

Cultures of Acinetobacter gyllenbergii 2P01AA were prepared from glycerol stocks by streaking onto LB agar plates. After incubation overnight at 32°C, isolated colonies were cultivated to stationary phase (OD₆₀₀ \sim 3.5) in LB at 32°C at 200 rpm (New Brunswick Scientific Innova 4200 incubator). A 1:100 inoculum of this culture was then used to prepare mid-log phase cultures (OD₆₀₀ \sim 1.5) that were collected by centrifugation at 5445g at 4°C for 10 min (Beckman Allegra 21R centrifuge), washed three times in PBS [or 50 mM HEPES (pH 7.5) containing 100 mM NaCl], and stored as cell pellets at -80° C. Native catalase was purified from A. gyllenbergii 2P01AA by a 3-step chromatographic procedure on a Bio-Rad DuoFlow 10 chromatography system, which was equipped with a conductivity monitor and UV optics module. Cell pellets were partially thawed, resuspended in Bugbuster (5 mL per gram of wet pellet), mixed with Protease Inhibitor Cocktail (10 μ L per 1 mL), and incubated at 22°C for 20 min. The suspension was clarified by centrifugation (5445g for 20 min at 4°C), the supernatant collected, and the soluble extract ($\sim 33.5 \text{ mL}$) was loaded at 5 mL/min onto a Bio-Rad UnoQ1 $(1.5 \times 11.3 \text{ cm})$ anion exchange column, which was equilibrated with 50 mM HEPES (pH 8.0). Elutions were performed at 5 mL/min with a NaCl step gradient in 50 mM HEPES (pH 8.0), which increased by 0.1 M NaCl every 25 mL from 0.2 to 0.8 M NaCl. Catalase activities (measured as described below) were detected in fractions eluting at ~ 24 mS/cm; these fractions were pooled and sequentially buffer exchanged and concentrated to 5-10 mL in 50 mM HEPES (pH 8.0) with an Amicon 10kD Ultra-15 Centrifugal Device (VWR). The concentrated sample was subjected to size-exclusion chromatography (SEC) by using a Hi-Prep Sephacryl S-300 High Resolution column $(1.6 \times 60 \text{ cm})$, running buffer of 50 mM HEPES (pH 8.0), and flow rate of 1 mL/min. The catalase-active fractions were pooled, adjusted to 1 M NaCl, and separated by hydrophobic interaction chromatography (HIC) with a Phenyl Sepharose 6 Fast Flow column $(1.6 \times 10 \text{ cm})$, which was equilibrated in 50 mM HEPES (pH 8.0) containing 1 M NaCl. Elution was performed at 1 mL/min by using a step gradient of 50 mM HEPES (pH 8.0), which decreased by 0.1 M NaCl every 35 mL. In sum, catalase activities were detected in the early fractions after SEC and in the fractions eluting between 0 and 0.1 M NaCl after HIC. Samples from each step in the purification were then analyzed by using 10% SDS-PAGE gels (Pierce Precise), which were run at 150 V for ~ 35 min (Bio-Rad Mini-Protean System), stained with Coomassie G-250, and quantified by densitometry (Un-Scan-It, Silk Scientific, Inc.).

2.3. Mass spectroscopy

Mass spectrometry was used to identify and characterize the catalase during and after the purification. Partially purified samples were separated on 10% SDS-PAGE gels, and a band corresponding to $\sim 60 \text{ kDa}$ was excised, subjected to in-gel trypsin digestion, and mass spectra of the products obtained on an Applied Biosystems 4800 MALDI/TOF mass spectrometer under MS and MS/MS modes (Institute for Integrated Research on Materials, Environment and Society, California State University, Long Beach). Spectra were assigned to bacterial sequences in the MSDB database by using 95% confidence intervals and the Mascot search engine (Matrix Science, Boston, MA). Upon purification, the amino acid sequence, intact molecular weight, subunit molecular weight, and ligand content were assessed. The catalase amino acid sequence was determined by liquid chromatographytandem mass spectrometry (LC-MS/MS) analyses (Agilent capillary HPLC, Thermo LTQ-FT Ultra mass spectrometer) on differing proteolytic digests. To obtain maximum coverage, several proteolytic reagents were used, including trypsin (Roche), LysC (Roche), GluC (Roche), subtilisin (Sigma-Aldrich), chymotrypsin (Roche), and formic acid for chemical lysis (EMD). In general, the digestion procedures included concentration of the sample to $> 50 \,\mu g/mL$, buffer exchange into 100 mM Tris-HCl (pH 8.0), addition of 0.1% RapiGest (Waters), and incubation at 99°C for 10 min. Upon conclusion and after cooling to room temperature (RT), 10 mM TCEP (Pierce) was added to the sample, incubated at RT for 30 min, after which 10 mM iodoacetamide (Sigma-Aldrich) was added, and the mixture incubated again for another 30 min at RT. Proteolytic enzymes (0.25-2 mg/mL) were then added to aliquots of the sample, incubated at 50°C for 1 h (25°C for chymotrypsin), and quenched by incubation in 0.1 M HCl at 37°C for 45 min. For the chemical lysis reactions, samples were incubated in 2% formic acid at 99°C for 120 min. The digests were then subjected to LC-MS/MS, and all spectra were matched to the NSCInr database with the MASCOT search engine and $\geq 95\%$ confidence intervals. Gaps in the coverage were treated by using error-tolerant searches, manual inspection of the raw spectral data, and sequence homology to catalases from other Acinetobacter ($\geq 95\%$ identity). In the homology alignments, amino acid identities were tentatively assigned only when 100% homology was observed across the aligned strains. The mass of the intact catalase was obtained on a traveling wave ion mobility quadrupole time-of-flight mass spectrometer (Waters Q-TOF Synapt G2 mass spectrometer) by using $> 250 \,\mu\text{g/mL}$ samples, which had been buffer exchanged into 100 mM NH₄OAc (pH 7.0) overnight at 4°C with a 7000 MWCO mini dialysis unit (Thermo-Pierce). Mass of the catalase subunit was obtained under denaturing conditions by using $>50 \,\mu$ g/mL samples and a time-of-flight mass spectrometer coupled to a nano-UPLC system equipped with a C18 nanospray column (Q-TOF Ultima & nanoAcquity; Waters, Milford, MA).

2.4. Absorption spectroscopy

Absorption spectra for the purified catalase were obtained with a Beckman Coulter DU-640 spectrophotometer. Spectra (240–800 nm) were acquired by using $16 \,\mu$ g/mL or 68 nM catalase in the absence and presence of 0.83 mM NaCN, where the mixtures were incubated for 10 min at 22°C prior to analysis.

2.5. Enzyme kinetics and inhibition

Enzyme kinetic studies on the purified catalase were performed by spectrally following the decomposition of H₂O₂ at 240 nm ($\varepsilon = 43.6 M^{-1} \text{ cm}^{-1}$). The change in absorbance was measured every 2s, for a minimum of 30s, on a Beckman Coulter DU- 640 spectrophotometer. All reactions were 1.00 mL in volume, run at RT (\sim 22°C), and thoroughly but gently mixed. Reaction rates were measured between 5 and 100 mM H₂O₂ in 50 mM HEPES (pH 7.5) in absence and presence of 100 or 200 mM NaCl. Reactions were initiated by addition of enzyme by using a final concentration of 67 ng/mL or 0.29 nM catalase, which was calculated by using the intact molecular weight obtained from mass spectral studies and an assumption of full iron loading. Kinetic constants were obtained by using rates obtained between 10 and $50 \text{ m}M \text{ H}_2\text{O}_2$, as regressions of the double-reciprocal plots within this concentration range provided coefficients of determination of $R^2 > 0.99$; the standard errors (se) of these regressions are reported herein. Inhibition studies were performed by using 3-amino-1,2,4-triazole (3AT), sodium azide (NaN₃), and sodium cyanide (NaCN). In summary, 6 µg/mL or 26 nM purified catalase was premixed with either $400 \,\mu M$ 3AT, 15 nM NaN₃, or 500 µM NaCN in 50 mM HEPES (pH 7.5), respectively. After incubation for 1 min at 22°C, the enzyme reaction rates were measured by using 60 ng/mL or 0.26 nM catalase and 20 mM H₂O₂ in 50 mM HEPES (pH 7.5). Control experiments on bovine liver catalase (BLC) were performed in parallel.

2.6. Temperature and pH stability profiles

Temperature stability profiles for catalase were obtained between the range of 10–100°C. Catalase samples (6 μ g/mL or 26 nM) were incubated at 5–10°C increments for 15 min, equilibrated to RT over ~ 30 min, and assayed for enzymatic activity. Reaction conditions included 60 ng/mL or 0.26 nM catalase, 20 mM H₂O₂, and 50 mM HEPES buffer (pH 7.5) in absence and presence of 100 mM NaCl. The pH stability profiles for catalase were also obtained. Assay conditions included 60 ng/mL or 0.26 nM catalase, and 20 mM H₂O₂, in 30 mM sodium pyrophosphate adjusted to pH 2.5–11.5; control experiments indicated no change in the buffered pH upon conclusion of the kinetic reactions. All other conditions are as described.

3. Results

3.1. Catalase purification

The native catalase from the H₂O₂-resistant Acinetobacter gyllenbergii 2P01AA was purified by a 3-step



FIG. 1. Electrophoretic analysis (10% SDS-PAGE) of the catalase purification: Bio-Rad molecular weight standard (lane 1); BLC control (lane 2); purified and concentrated catalase from *A. gyllenbergii* 2P01AA (lane 3); catalase sample after anion exchange (lane 4); catalase sample after SEC (lane 5); catalase sample after hydrophobic interaction exchange (lane 6); and BLC control (lane 7).

chromatographic procedure following the order of anion exchange, SEC, and HIC (Fig. 1). As indicated in Table 1, the overall purification yield of the native catalase was 36%, which correlated to a \sim 150-fold enrichment in specific activity to yield a final value of 8.7×10^6 Units/mg. Densitometry of the catalase samples separated by using 10% SDS-PAGE gels indicated a >95% purity after the 3-step procedure. Initial experiments showed that the order of separation was significant, as separations following the sequence of anion exchange, HIC, and SEC provided only $\sim 40\%$ purity. The identity of the purified protein was confirmed as catalase by mass spectral analyses of trypsin digests of (A) a $\sim 60 \,\mathrm{kDa}$ protein excised from electrophoretic gels during the purification, which provided matches to catalase from Acinetobacter calcoaceticus (O83WC7 ACICA), and (B) the purified sample, which provided matches to Acinetobacter junii SH205 (D0SPS7).

3.2. Sequence and structural homology models

The amino acid sequence of catalase from *A. gyllenbergii* 2P01AA was determined through in-depth peptide mapping and the use of multiple proteolytic reagents, which ultimately provided 96% sequence coverage (Fig. 2). Error-tolerant searches of the mass spectra confirmed the identity of Ser³⁸⁵, which corresponded to Gly in other *Acinetobacter*, and the constructed 1° structure revealed that the N-terminal methionine had been cleaved. BLAST comparisons in the NCBI database denoted a clade III catalase corresponding to the katE gene (von Ossowski *et al.*, 1991). Distance trees (NCBI) placed the catalase into a sub-branch of the *Acinetobacter* populated by *A. junii* and *A. gyllenbergii*, and indicated that the catalase was more closely related to *A. junii*,

despite the confirmed taxonomy of the microorganism as A. gyllenbergii. The unassigned amino acids in the sequence (4% of the coverage) were assigned by sequence homology to catalases from 12 different species of Acinetobacter, where each catalase shared high identity values of $\geq 95\%$ with the target sequence. At each position, the alignments showed 100% homology across all species with the exception of X^{274} (Fig. 2), which varied between Asp and Glu. For demonstrative purposes, alignments of the target sequence to A. junii SH205 (D0SPS7) and A. gyllenbergii MTCC 11365 (S3YMH9) are provided in Fig. 3A, which together aptly show the 100% homology at the unassigned positions. To remain consistent within the sequence analysis, the Nterminal amino acid for the purified catalase is assumed to be the cleaved methionine residue. Additionally, alignments across all known strains of A. gyllenbergii, as illustrated by the alignment to A. gyllenbergii MTCC 11365 (Fig. 3A), revealed several mutations that were unique to the 2P01AA strain. These mutations included V^{331} (A³³¹ in *A. gyllenbergii* MTCC 11365), I³⁷⁹ (T³⁷⁹ in *A. gyllenbergii* MTCC 11365), Ser³⁸⁵ (G³⁸⁵ in *A. gyllenbergii* MTCC 11365), and the C-terminal residue of I⁵⁰⁶ (L⁵⁰⁶ in *A. gyllenbergii* MTCC 260 11365).

A structural homology model for catalase (Fig. 3B) was built by using the SWISS-MODEL template library (Arnold *et al.*, 2006; Benkert *et al.*, 2011; Biasini *et al.*, 2014), where BLAST and HHBlits searches of the target sequence identified the closest evolutionary related structure (from the Protein Data Bank) to be catalase (2isa) from *Vibrio salmonicida* (sequence identity, 74.16%). The ligand-free structural homology model displayed several aspects typical to small subunit catalases, including an intertwining Nterminal arm, an antiparallel eight-stranded β -barrel, and a

TABLE 1. CATALASE PURIFICATION YIELDS FROM A. GYLLENBERGII 2P01AA

Purification step	Total protein (mg)	Total activity (Units)	Specific activity (Units/mg)	Purification fold	Yield (%)
Crude extract Anion exchange Size exclusion	70 8.0 3.0	4.0×10^{6} 2.2×10^{6} 2.0×10^{6}	5.8×10^4 2.7 × 10 ⁵ 6.9 × 10 ⁵	4.7 12	54% 49%
Hydrophobic interaction	0.17	1.4×10^{6}	8.7×10^{6}	150	36%

10	20	30	40	50	60
MSQDDKKCPY	SHLTTDFGAP	VVDNQNSMTA	GARGPLLAQD	LWLNEKLANF	VREVIPERRM
70	80	90	100	110	120
HAKGSGAFGT	FTVTHDITQY	TRAKIFSEIG	KKTEMFARFT	TVAGERGAAD	AERDIRGFAL
130	140	150	160	170	180
KFYTEEGNWD	MVGNNTPVFF	LRDPRKFPDL	NKAVKRDPKT	NLRSATNNWD	FWTLLPEALH
190	200	210	220	230	240
QVTIVMSDRG	IPASYR HMHG	FSSHTY SFIN	AANERFW VKF	HFRTQQGIKN	LTDAEAGELV
250	260	270	280	290	300
GQDRESHQRD	LFDAIERQDY	PKWTLFVQIM	PEQX AEKVPY	HPFDLTKVWP	HGDYPLIEVG
310	320	330	340	350	360
EFELNRNSEN	FFLDVEQSAF	APSNLVPGIS	VSPDRMLQAR	LFNYADAQRY	RLGVNYQQIP
370	380	390	400	410	420
VNAARCPVHS	NHRDGQGRID	ANYGSLPHYE	PNSFGQWQEQ	PQYKEPPLKI	NGDADFWDYR
430	440	- 450	460	470	480
EDDNDYFSQP	RALFELMTPE	QQDALFGNTA	RAMGDALDFI	KYRHIRNCYA	CHPAYGEGVA
490	500				
KALGMTVADA	QAARETDPAR	HLPSFI			

FIG. 2. Amino acid sequence of catalase from *A. gyllenbergii* 2P01AA, where the residues identified by mass spectrometry are in bold font, while those assigned by homology are in nonbold font; the underlined amino acid (S³⁸⁵) was confirmed with error-tolerant searches, X²⁷⁴ represents either Asp or Glu, the active site residues (His⁶¹, Asn¹³⁴, Asn¹³⁵, Arg³⁴⁰, and Tyr³⁴⁴) are highlighted by boxes, and the sequence homology studies (UniProt) utilized alignments to catalases from *A. junii* MTCC 11364 (S7Y7D6), *A. junii* CIP 107470 (N9AZS1), *A. junii* SH205 (D0SPS7), *A. junii* CIP 64.5 (N9CBX8), *Acinetobacter* sp. NIPH 236 (N8R283), *Acinetobacter* sp. NIPH 284 (N9K5H2), *Acinetobacter* sp. NIPH 809 (N8QMK9), *A. gyllenbergii* MTCC 11365 (S3YMH9), *A. gyllenbergii* CIP 10306 (S3N7M0), *Acinetobacter* sp. ANC 3862 (N9N9I9), *A. gyllenbergii* NIPH 230 (V2TAZ6), and *Acinetobacter* sp. CIP 70.18 (N9RVW3).

C-terminal α -helical domain. Analysis of the structural model revealed that the mutations unique to the 2P01AA strain, among the known catalases from *A. gyllenbergii* (apart from the C-terminal residue), were all found at or near the interfaces between the catalase subunits, which may relate to the enhanced stability and kinetic properties. Further, Tyr³⁴⁴ was identified as the potential proximal Fe ligand, while His⁶¹, Asn¹³⁴, and Arg³⁴⁰ comprised the major determinants of the active site. In summary, these results support the purification of a clade III catalase, which lacks an N-terminal methionine, contains a total of 505 amino acids per subunit, and displays a theoretical subunit pI of 5.81.

3.3. Mass and absorption spectroscopy

Mass spectrometry on the denatured sample provided a resolved mass of 57,813 Da for the catalase subunit, which is within experimental error of the theoretical molecular weight $(57,796 \text{ Da}, \text{ when including } D^{274})$. Further, analyses on the native samples provided a resolved mass of 234,613 Da for the intact catalase, which supported the purification of a catalase tetramer containing 4 Fe-heme (heme b) and 1-2 NADPH ligands per catalase. Absorption spectroscopy (Fig. 4) confirmed the presence of Fe-heme with a Soret band at 403 nm (molar absorptivity of $\sim 1.6 \times 10^6 M^{-1} \text{ cm}^{-1}$) and heme charge transfer bands at 502 and 625 nm (Deisseroth and Dounce, 1970; Obinger et al., 1997; Thompson et al., 2003). As shown in Fig. 4, the absorbance spectra were sensitive to the addition of cyanide, which yielded (A) a red shift in the λ_{max} of the Soret feature from 403 to 424 nm, (B) loss of peaks at 502 and 625 nm, and (C) the gain of the α and β -peaks at 550 and 585 nm. Together, these features were consistent with conversion to a 6-coordinate low spin center upon cyanide ligation. Thus, the combined characterizations supported the purification of a Fe-hemecontaining, small subunit, and tetrameric catalase with the native state possessing 5-coordinate high spin ferric centers.

3.4. Enzyme kinetics and inhibition

Kinetic studies on the catalase from A. gyllenbergii 2P01AA (Fig. 5) revealed significant substrate inhibition at concentrations of $>50 \text{ mM} \text{ H}_2\text{O}_2$ in 100 mM NaCl and $>10 \text{ mM} \text{ H}_2\text{O}_2$ in 200 mM NaCl [in 50 mM HEPES (pH 7.5)]. This non-Michaelis-Menten behavior is common to catalases and accordingly treated by including only lower H_2O_2 concentrations in the analyses (Switala and Loewen, 2002). Apparent kinetics constants were thus obtained by using the rates collected between 10 and $50 \text{ m}M \text{ H}_2\text{O}_2$ in the absence and presence of 100 mM NaCl (Fig. 5). This analysis yielded an apparent $K_{\rm m}$ of $110 \,{\rm m}M$ (se = 29 mM), which increased to 230 mM (se = 110 mM) in the presence of 100 mMNaCl [in 50 mM HEPES (pH 7.5)]. The V_{max} for catalase was $1.1 \times 10^{10} \ \mu M \ s^{-1} \ g^{-1} \ (se = 0.3 \times 10^{10} \ \mu M \ s^{-1} \ g^{-1})$, which remained relatively constant in 100 mM NaCl at 1.2×10^{10} $\mu M \, \mathrm{s}^{-1} \, \mathrm{g}^{-1}$ (se = 0.6 × 10¹⁰ $\mu M \, \mathrm{s}^{-1} \, \mathrm{g}^{-1}$). The corresponding apparent k_{cat} values were 2.5 × 10⁶ s⁻¹ (se = 0.6 × 10⁶) and 2.9 × 10⁶ s⁻¹ (se = 1.4 × 10⁶), respectively. Lastly, the apparent $k_{\mathrm{cat}}/K_{\mathrm{m}}$ was 2.2 × 10⁷ s⁻¹ M^{-1} (se = 0.09 × 10⁷), which decreased to $1.2 \times 10^7 \text{ s}^{-1} M^{-1} (se = 0.05 \times 10^7)$ in 100 mM NaCl. Together, the parameters indicate that the presence of NaCl decreased the rate of substrate capture (k_{cat}/K_m) yet caused no significant impact on the rate of product release (k_{cat}) (Mogul and Holman, 2001). For demonstrative



FIG. 3. Sequence and structural homology models for the catalase from *A. gyllenbergii* 2P01AA, where (**A**) the alignments (COBALT, NCBI) against the catalases from *A. junii* SH205 (D0SPS7) and *A. gyllenbergii* MTCC 11365 (S3YMH9) are signified by the respective abbreviations of 2P01AA, SH205, MTCC 11365; and (**B**) the structural homology models (SWISS-MODEL) of the catalase subunit, tetramer assembly, and surface view of the tetramer assembly display the relative positions of the residues (black, space-filling format, or ovals) that differ among the *A. gyllenbergii* strains. For the alignments, the unassigned amino acids are denoted by X, the asterisk demarcation (*) represents a fully conserved amino acid, the colon demarcation (:) represents conservation among the amino acid group, the period demarcation (.) represents conservation of the general amino acid property, and the boxes highlight the amino acids discussed in the narrative.



FIG. 4. Absorbance spectra of catalase from *A. gyllenbergii* 2P01AA in the absence (black line) and presence (gray line) of cyanide, where the impact of cyanide on the Soret bands (403 and 424 nm), heme charge transfer bands (502 and 625 nm), and α and β -bands (550 and 585 nm) are consistent with a native catalase that contains a 5-coordinate high spin ferric center.



FIG. 5. Actual and theoretical impacts of H_2O_2 concentration on the rates of catalase, where the experimentally determined rates were obtained by using 5–100 mM H_2O_2 in 50 mM HEPES (pH 7.5) containing 0 (circles), 100 (diamonds), and 200 (squares) mM NaCl; the theoretical catalase rates (dashed lines) were calculated with the apparent kinetic constants and a Michaelis-Menten kinetic model, which excludes substrate inhibition (all rates were measured in triplicate, error bars represent the standard deviation, and all regressions were performed using coefficients of determination of ≥ 0.99).

purposes, the theoretical enzyme rates for catalase were calculated with the apparent kinetic constants and are displayed in Fig. 5. These comparisons of actual and theoretical rates effectively illustrate the pronounced substrate and solute inhibition by H_2O_2 and NaCl, respectively. Consequently, the rate data obtained in 200 mM NaCl could not be mathematically fitted due to significant substrate inhibition.

For control purposes, kinetic analyses were also performed on BLC. Under our conditions, significant substrate inhibition was observed at $>25 \text{ m}M \text{ H}_2\text{O}_2$; therefore, only the rates obtained between 10 and 20 mM H₂O were used in the kinetics. These analyses yielded apparent $K_{\rm m}$ values of 18 and 20 mM in the absence and presence of 100 mM NaCl, respectively. For comparative purposes, BLC was assumed to be fully loaded with iron; hence, an apparent k_{cat} of $7.4 \times 10^4 \text{ s}^{-1}$ was obtained, which decreased to $4.6 \times 10^4 \text{ s}^{-1}$ in 100 mM NaCl. The apparent k_{cal}/K_m was 3.6×10^6 s⁻¹ M^{-1} , which decreased to 1.9×10^6 s⁻¹ M^{-1} in 100 mM NaCl. Therefore, the presence of NaCl impacted both the rate of substrate capture and product release for BLC, which is in contrast to measured impact on the purified Acinetobacter catalase. Together, the impact of NaCl on the kinetics is consistent with a competitive-type inhibition by NaCl for the purified catalase, where increasing NaCl concentration correlates with an increasing $K_{\rm m}$ and relatively constant V_{max} . In comparison, the kinetics of BLC are consistent with noncompetitive inhibition, where increasing NaCl concentration correlates with a relatively constant $K_{\rm m}$ and decreasing V_{max} . However, these kinetic outcomes are most likely due to the impacts of ionic strength on the solution dynamics, protein structure, and protein ionization (Nakamura and Kimura, 1971; Loo and Erman, 1975; Luo et al., 2004). Nevertheless, on average, as displayed in Fig. 6, NaCl reduced the rates of both enzymes by $51 \pm 4\%$ between 10 and 25 mM H_2O_2 . However, at 30 mM H_2O_2 , the rate of BLC was substantially decreased by ~85%, whereas no

change was observed for the catalase from the 2P01AA strain. Hence, when compared to BLC, the purified catalase possessed a greater tolerance toward substrate inhibition in 100 mM NaCl. This tolerance is further demonstrated by comparison of the rates in 200 mM NaCl, where no activity was measured for BLC. Lastly, inhibition studies in which 3AT, NaN₃, and NaCN were used showed $10.1\pm1.1\%$, $16.5\pm2.5\%$, and $88.4\pm1.1\%$ (standard deviation) reductions in rate, respectively.



FIG. 6. Comparison of catalase inhibition by NaCl at differing concentrations of H_2O_2 for *A. gyllenbergii* 2P01AA (black marked line and circles) and BLC (gray marked line and squares), where the difference in rates measured in the absence (v_0) and presence (v_0^{NaCl}) of 100 mM NaCl between 10 and 30 mM H_2O_2 [in 50 mM HEPES (pH 7.5)] is expressed as a percent of the activity remaining after inhibition [Percent Activity (%) = (v_0^{NaCl}/v_0) × 100] (all rates were measured in triplicate, and error bars represent the standard deviation).



FIG. 7. Impacts of (**A**) temperature on catalase stability, which was assayed by measuring the change in catalase activity after 15 min incubations at the respective temperature, and (**B**) pH on catalase activity for *A. gyllenbergii* 2P01AA (black marked line and circles) and BLC (gray marked line and squares), where the activity assays were conducted by using 30 mM pyrophosphate buffers adjusted to pH values between 2.5 and 11.5 [assay conditions utilized 20 mM H₂O₂ in 50 mM HEPES (pH 7.5), all rates were measured in triplicate, and error bars represent the standard deviation].

3.5. Temperature and pH stability

Temperature and pH stability studies were conducted on catalase. Temperature stability provided an apparent $T_{\rm m}$ of 56°C and indicated that the catalase retained $\sim 7\%$ of the activity after incubation at 60°C for 15 min and that \geq 70°C abolished all activity (Fig. 7A). Addition of NaCl had no major impact on the thermal stability of the enzymatic activity. The $T_{\rm m}$ for BLC has also been shown to be 56°C, with 100% deactivation occurring at ~ 62° C (Switala *et al.*, 1999). The catalase enzyme also was active from pH 4 to 11.5 and displayed a pH optimum of 5–11.5 (Fig. 7B), which is among the broadest pH optima for catalases (Esaka and Asahi, 1982; Allgood and Perry, 1986; Fu et al., 2014). Notably, no loss in activity was measured between pH 5 and 11.5. However, at the more acidic values of pH 4 and 2.5, 50% and 100% losses in activity were obtained, respectively. In comparison, for BLC, the range in pH optima was 6–9, with significant losses at pH 5 (\sim 55%) and pH 10.5 (\sim 45%) and complete loss of activity at pH values of 2.5 and 11.5, respectively.

4. Discussion

Catalase from Acinetobacter gyllenbergii 2P01AA was purified with a 3-step chromatographic procedure, ultimately providing a 36% yield, final specific activity of 8.7×10^6 Units/mg, and purity of >95%. The resolved masses for the native catalase and denatured subunit were 234,613 and 57,813 Da, respectively, thereby supporting the purification of a tetramer catalase containing 4 Fe-heme b ligands. Absorbance spectra indicate that the Fe-heme groups were 5coordinate high spin ferric centers. Sequence alignments indicate a clade III catalase with 505 amino acids per expressed and processed subunit, whereas the structural homology model suggests that mutations along the subunit interfaces were responsible for providing the unique stability and kinetic parameters among the catalases from differing strains of A. gyllenbergii. Kinetic studies provided an apparent $K_{\rm m}$ of 110 mM, which is similar to those displayed by catalases from Helicobacter pylori, Listeria seeligeri, Saccharomyces cerevisiae but ~2-fold lower than that of Bac*teroides fragilis* and *Serratia marcescens* and ~4-fold lower than that of *Aspergillus niger* and *Proteus mirabilis* (Switala and Loewen, 2002). Moreover, the respective k_{cat} and k_{cat}/K_m values of 2.5×10^6 s⁻¹ and 2.2×10^7 s⁻¹ M^{-1} are among the highest reported for bacterial catalases (Switala and Loewen, 2002; Thompson *et al.*, 2003).

Catalase was inhibited by 3AT, NaN₃, NaCN; displayed a competitive-type inhibition by NaCl; and was significantly inhibited by substrate >50 mM H₂O₂. In comparison to BLC, the catalase from the 2P01AA strain was ~3-fold more active at 30 mM H₂O₂ in 100 mM NaCl. The catalase was also more tolerant toward NaCl concentration, as moderate activity at 200 mM NaCl was detected, while BLC displayed irreproducible rates. The catalase was moderately thermostable, where slight, but reproducible, activities were measured at 60°C, whereas for BLC the activity is abolished at 62°C (Switala *et al.*, 1999). The catalase was also very pH stable with no measurable loss in activity between pH 5 and 11.5. Thus, our combined characterizations support the purification of small-subunit, extremely pH stable, alkalitolerant, halotolerant, and moderately thermostable catalase.

Together, these properties are consistent with the survival features of A. gyllenbergii 2P01AA, which is extremotolerant toward H₂O₂ and was isolated from the oligotrophic and lowhumidity spacecraft assembly facilities. Survival studies in nutrient-rich media show that A. gyllenbergii 2P01AA displays no loss in survival at 100 mM H₂O₂ and only a 2-log reduction in $330 \text{ mM H}_2\text{O}_2$ (Derecho *et al.*, 2014), which represents perhaps the highest survival against H_2O_2 among Gram-negative and non-spore-forming bacteria (McCoy et al., 2012; Derecho et al., 2014). These trends are consistent with the apparent $K_{\rm m}$ values, which indicate that sufficient catalase activities would be found within the cell in the presence of 100 and 330 mM H₂O₂. For example, the catalase enzyme at these H₂O₂ concentrations would not be saturated and accordingly would degrade H₂O₂ at rates of 30% and 59% of V_{max} —which assumes (A) a relatively high intracellular NaCl concentration of 100 mM (Schultz et al., 1962), (B) an apparent $K_{\rm m}$ of $230 \,\mathrm{m}M$ in $100 \,\mathrm{m}M$ NaCl, and (C) a substantially decreased substrate inactivation/inhibition in vivo. Relatedly, the extremely high catalase specific activities in extracts of A. gyllenbergii 2P01AA (1800±350 Units/mg) (Derecho et al., 2014) are likely due to the high apparent k_{cat} and k_{cat}/K_m values, and not necessarily due to high catalase abundances, as changes in catalase volumes upon H_2O_2 exposure have yet to be detected by proteomics. Further, initial experiments showed that cultures of spacecraftassociated Acinetobacter grown in nutrient-rich and nutrientlimited media possess similar total catalase specific activities, which suggests that the expression of catalase is relatively independent of carbon nutrient content (unpublished results). Thus, the measured alkali-tolerance of catalase is potentially significant, as A. gyllenbergii 2P01AA was initially retrieved from the floor of the Mars Phoenix assembly facility, which was routinely cleaned with alkaline detergents, and subsequently isolated on R2A agar plates, which were adjusted to pH 11 (Vaishampayan et al., 2010; Derecho et al., 2014). Further, the halotolerance and increase in $K_{\rm m}$ under high salt concentrations are consistent with survival in low-humidity environments (Potts, 1994) and thereby potentially relate to survival in the low-humidity assembly facilities.

5. Conclusion

In conclusion, our work reveals that the catalase kinetic and stability properties are consistent with molecular adaptations toward the alkaline, low-humidity, and potentially oxidizing conditions of the spacecraft assembly facilities (NASA-KSC, 1999; Vaishampayan et al., 2012; Derecho et al., 2014). To date, there remain very few reports that focus on the proteomic underpinnings of the extremotolerance of spacecraft-associated microorganisms (Checinska et al., 2012; McCoy et al., 2012; Derecho et al., 2014). Hence, this study reveals key enzymatic details into the survival of A. gyllenbergii 2P01AA and represents perhaps the first purification and characterization of a protein from a spacecraft-associated microorganism. Thus, our collective results support the hypothesis that the selective pressures of the spacecraft assembly conditions impact the microbial communities at the molecular level, which may have broad implications for future life-detection missions.

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Abbreviations Used

BLC = bovine liver catalase HIC = hydrophobic interaction chromatography LB = lysogeny broth LC-MS/MS = liquid chromatography-tandem mass spectrometry OTUs = operational taxonomic units PBS = phosphate-buffered saline RT = room temperature se = standard error SEC = size-exclusion chromatography