

Decarbonylation of Ketones with Bidentate *N*-Heterocyclic Carbene Nickel Complexes

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What is decarbonylation?

- Removing CO (carbonyl) from a molecule
- Significant for pharmaceutical and fine chemical production
- Alternative route for synthesizing carbon-carbon bonds



- Difficult to achieve
- First decarbonylation of ketones with nickel recently reported

Goals

- Synthesize new nickel catalysts
- Test nickel catalysts in decarbonylation
- Determine effect of ligand sterics in *decarbonylation*



Decarbonylation of ketones



Chemical Formula: C₁₃H₁₀O₃ Exact Mass: 214.06 Molecular Weight: 214.22 m/z: 214.06 (100.0%), 215.07 (14.1%) Elemental Analysis: C, 72.89; H, 4.71; O, 22.41 Chemical Formula: C₁₂H₁₀ Exact Mass: 154.08 Molecular Weight: 154.21 m/z: 154.08 (100.0%), 155.08 (13.0%) Elemental Analysis: C, 93.46; H, 6.54







Preparation of ligand precursors





Decarbonylation Reactions with other Ketones

R group	[Ni] catalyst	Product identified?
-H	<i>t</i> -butyl	Yes
-H	Mes	Νο
-NCCEt ₂	<i>t</i> -butyl	Νο
-NEt ₂	Mes	Yes
-OCH ₃	<i>t</i> -butyl	Νο
-OCH ₃	Mes	Νο
-OH	<i>t</i> -butyl	Νο
-OH	Mes	Yes
-CH ₃	<i>t</i> -butyl	Yes
-CH ₃	Mes	Yes

Conclusions and Directions

- **Initial decarbonylation reactions demonstrated**
- Utilize preliminary data to improve conversion



Expand scope to additional functionalized ketones

Develop new nickel catalysts

References

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