

## RESPeCT Summer Institute Professional Development Leader Guide (PDLG)

<b>Grade Level</b>	2	<b>Day</b>	3	<b>STeLLA Strategy</b>	STL Strategy 4: Analyze and Interpret Data and Observations STL Strategy 5: Construct Explanations and Arguments	<b>Subject Matter Focus</b>	Properties of Matter
<b>Focus Questions</b>	<ul style="list-style-type: none"> <li>• How can analyzing data and constructing explanations help students <i>move forward</i> toward deeper understandings of science ideas?</li> <li>• Are physical changes reversible at room temperature? What is your evidence?</li> <li>• Are atoms created or destroyed when matter undergoes a physical or chemical change? How can you prove your ideas?</li> </ul>						
<b>Main Learning Goals</b>	<p>Participants will understand the following:</p> <ul style="list-style-type: none"> <li>• In addition to challenge questions, the Student Thinking Lens (STL) strategies include activities that move student thinking forward toward more-scientific understandings.</li> <li>• STL strategies 4 and 5 are two activities that can be used to move student thinking forward: Engage students in analyzing and interpreting data and observations (strategy 4), and engage students in constructing explanations and arguments (strategy 5).</li> <li>• Analyzing and interpreting go beyond making observations to organizing data, identifying patterns and looking for meaning in the data, and searching for relationships between science ideas and data.</li> <li>• Constructing explanations involves making a claim, supporting the claim with evidence and reasoning, and coming up with alternatives that challenge the claim (argumentation).</li> <li>• Atoms are neither created nor destroyed when matter undergoes a physical or chemical change. The number of atoms remains the same before and after a change.</li> <li>• The phenomena of matter can be explained by answering the central questions, <i>What is matter made of? How can matter change?</i></li> </ul>						
<b>Preparation</b>				<b>Materials</b>			<b>Videos</b>
<p><b>Daily Setup Tasks</b></p> <ul style="list-style-type: none"> <li>• Check that video clips are correctly linked to PowerPoint (PPT) slides.</li> <li>• Set up PowerPoint.</li> <li>• Make sure video clips play correctly with good sound.</li> <li>• Arrange furniture and food.</li> <li>• Arrange participant materials.</li> <li>• Put up posters and charts.</li> </ul> <p><b>Planning and Preparation Tasks</b></p> <ul style="list-style-type: none"> <li>• Study the PDLG, PowerPoint slides (PPTs),</li> </ul>				<p><b>Posters and Charts</b></p> <ul style="list-style-type: none"> <li>• STeLLA Framework and Strategies poster</li> <li>• Day-3 Agenda (chart)</li> <li>• Day-3 Focus Questions (chart)</li> <li>• Norms for Working Together (chart)</li> <li>• Effective Science Teaching chart (from day 1)</li> <li>• Strategy charts from days 1 and 2 (STL strategies 1–3)</li> <li>• Parking Lot poster</li> </ul> <p><b>Handouts in RESPeCT PD Binder Front Pocket</b></p>			<ul style="list-style-type: none"> <li>• <a href="#">Video Clip 3.1</a>: Griffin classroom (analyze and interpret, strategy 4); 3.1_mspcp_gr.2_matter_griffin_L2_c4–c6</li> <li>• <a href="#">Video Clip 3.2</a>: Fowler classroom (construct explanations and arguments, strategy 5); 3.2_mspcp_gr.2_matter_fowler_L5_c7–c9</li> </ul>

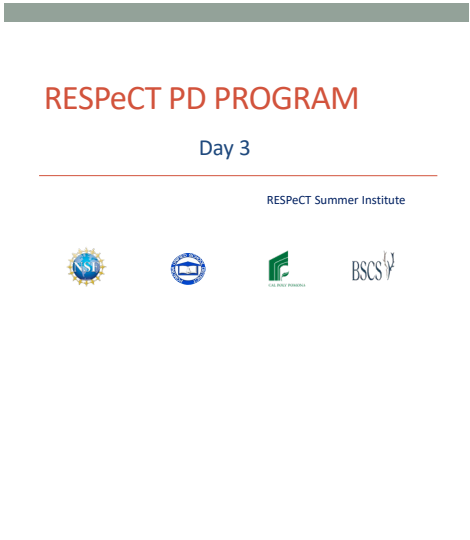
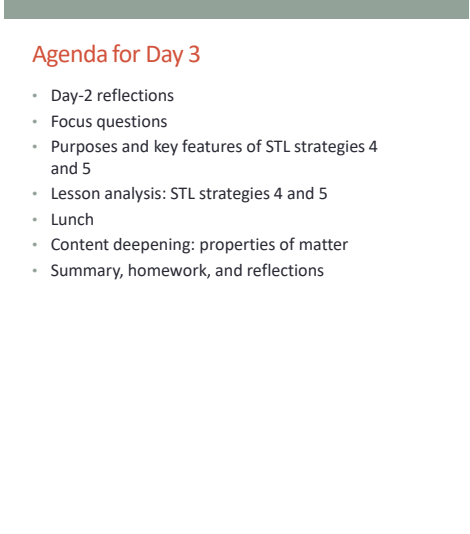
<p>video clips, and handouts. Make changes to PPTs if needed.</p> <ul style="list-style-type: none"> <li>• Review the reflections from day 2 and create a summary slide.</li> <li>• Watch video clips and anticipate participant responses.</li> <li>• Prepare charts for the day’s agenda and focus questions.</li> <li>• As needed, review the activities from Properties of Matter lessons 1a, 2a, and 7 in the lesson plans binder. Some activities have been modified slightly from the original lessons.</li> <li>• For content deepening: <ul style="list-style-type: none"> <li>• For the melting/freezing demonstration, set up the griddle and beakers before the lesson begins, making sure to place it in a location where no one will trip over the electrical cord or accidentally knock over the beakers. Turn on the griddle (low to medium heat) at the beginning of the lesson to let it warm up. Don’t place the beakers of chocolate, crayons, and butter on the griddle until the demo begins.</li> <li>• For the physical-changes experiment, place a bottle of water in the freezer overnight. You may want to freeze an extra bottle as a replacement if needed. Make sure to fill the freezer bottle and the room-temperature bottle with the same amount of water. The water in the freezer bottle will expand overnight, but both bottles should weigh the same.</li> </ul> </li> </ul>	<ul style="list-style-type: none"> <li>• Z-fold summary chart: Student Thinking Lens Strategies</li> </ul> <p><b>Handouts in RESPeCT PD Binder, Day 3</b></p> <ul style="list-style-type: none"> <li>• 3.1 Quick Reference Tools for Strategies 4 and 5</li> <li>• 3.2 Practice Identifying Strategies 4 and 5 in Student Work</li> <li>• 3.3 Transcript for Video Clip 3.1</li> <li>• 3.4 Transcript for Video Clip 3.2</li> <li>• 3.5 Daily Reflections—Day 3</li> </ul> <p><b>PD Leader Masters, Days 1–4</b></p> <ul style="list-style-type: none"> <li>• PD Leader Master: Practice Identifying Strategies 4 and 5 in Student Work</li> </ul> <p><b>Supplies</b></p> <ul style="list-style-type: none"> <li>• Science notebooks</li> <li>• Chart paper and markers</li> <li>• Lesson materials kits</li> <li>• For melting/freezing demonstration (adapted from lessons 1a and 2a): <ul style="list-style-type: none"> <li>• Small griddle</li> <li>• 3 clear-glass, heat-resistant beakers</li> <li>• 3 wooden craft sticks (for stirring)</li> <li>• Pot holders</li> <li>• Chocolate chips or pieces (milk chocolate)</li> <li>• Crayons or crayon pieces (one color)</li> <li>• Butter</li> </ul> </li> <li>• For physical-changes experiment (adapted from lesson 7): <ul style="list-style-type: none"> <li>• 2 balance scales (1 per team)</li> <li>• 24 white (hydrogen) Legos and 12 red (oxygen) Legos (for Team 1)</li> <li>• 1 bottle of frozen water and 1 bottle of room-temperature water (for Team 2)</li> </ul> </li> <li>• For chemical-changes experiment (adapted from lesson 7): <ul style="list-style-type: none"> <li>• 2 balance scales (1 per team)</li> <li>• 1 lime-green (sodium) Lego, 5 white (hydrogen) Legos, 5 red (oxygen) Legos, 3 black (carbon) Legos (1 set per team)</li> </ul> </li> </ul>	
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	<ul style="list-style-type: none"> <li>• Bag containing baking soda and vinegar (1 per team):             <ul style="list-style-type: none"> <li>• 1 quart-sized, sealable, plastic freezer bag</li> <li>• 1 teaspoon of baking soda</li> <li>• 2 tablespoons of white vinegar</li> <li>• Clear plastic vial with snap-on cap</li> <li>• Paper towels</li> </ul> </li> </ul> <p><b>PD Resources</b></p> <ul style="list-style-type: none"> <li>• STeLLA strategies booklet</li> <li>• RESPeCT PD program binder</li> <li>• RESPeCT lesson plans binder</li> </ul> <p><b>Resources in Lesson Plans Binder</b></p> <p><i>Resources section:</i></p> <ul style="list-style-type: none"> <li>• Properties of Matter Content Background Document</li> <li>• Common Student Ideas about Properties of Matter</li> </ul>	
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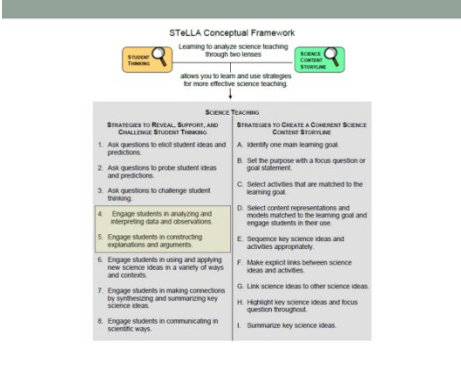
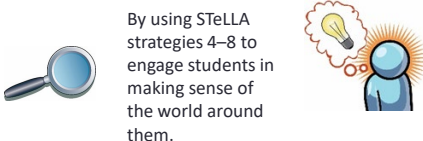
### DAY 3 SESSION OUTLINE

Time	Activities	Purpose
8:00–8:35 35 min	<b>Getting Started: Housekeeping, Agenda, Day-2 Reflections, Focus Questions, STL Strategies</b>	<ul style="list-style-type: none"> <li>• Build community by sharing participants’ reflections from day 2.</li> <li>• Set the stage for a day of learning.</li> <li>• Emphasize the theme for the rest of the week: What do we do with the ideas we’ve elicited from students? How do we help them change and advance their understandings of science concepts?</li> </ul>
8:35–9:35 60 min	<b>Introducing Student Thinking Lens (STL) Strategies 4 and 5</b>	<ul style="list-style-type: none"> <li>• Develop an initial understanding of strategy 4: Engage students in analyzing and interpreting data and observations.</li> <li>• Develop an initial understanding of strategy 5: Engage students in constructing explanations and arguments.</li> <li>• Examine the relationships among the science practices of observing, analyzing and interpreting, and constructing explanations and arguments.</li> </ul>
9:35–12:00 145 min (Includes 10-min break)	<b>Lesson Analysis: STL Strategies 4 and 5</b>	<ul style="list-style-type: none"> <li>• Use lesson analysis of classroom videos to better understand strategies 4 and 5, how they’re related, and how they can challenge student thinking to move forward.</li> <li>• Deepen science-content knowledge of properties of matter through lesson analysis.</li> </ul>
12:00–12:45 45 min	<b>LUNCH</b>	
12:45–3:15 150 min (Includes 10-min break)	<b>Content Deepening: Properties of Matter</b>	<ul style="list-style-type: none"> <li>• Deepen participants’ understandings of what matter is made of and how it can change.</li> <li>• Deepen participants’ science-content knowledge by conducting investigations from the Properties of Matter lessons.</li> </ul>
3:15–3:30 15 min	<b>Wrap-Up: Summary, Homework, and Reflections</b>	<ul style="list-style-type: none"> <li>• Reflect on the day’s learning and summarize key ideas about the science content and strategies 4 and 5, linking those ideas to participants’ images of effective science teaching and changes they want to make in their individual teaching practices.</li> </ul>

**DAY 3**

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
<p>8:00–8:35 35 min</p> <p><b>Getting Started</b></p> <p>Slides 1–8</p>	<p><b>Purpose</b></p> <ul style="list-style-type: none"> <li>• Build community by sharing participants’ reflections from day 2.</li> <li>• Set the stage for a day of learning.</li> <li>• Emphasize the theme for the rest of the week: What do we do with the ideas we’ve elicited from students? How do we help them change and advance their understandings of science concepts?</li> </ul> <p><b>Content</b></p> <ul style="list-style-type: none"> <li>• Student Thinking Lens (STL) strategies reveal student thinking (elicit and probe strategies) and challenge student thinking (the rest of the strategies).</li> <li>• STL strategies are divided into questions (elicit, probe, and challenge) and activities.</li> <li>• A variety of strategies can be used to move student thinking forward. Today’s focus is STL strategy 4 (Engage students in analyzing and interpreting data and observations) and strategy 5 (Engage students in constructing explanations and arguments).</li> </ul> <p><b>What Participants Do</b></p> <ul style="list-style-type: none"> <li>• Discuss the reflections from day 2.</li> <li>• Listen to an overview of the</li> </ul>	 <p>RESPeCT PD PROGRAM</p> <p>Day 3</p> <p>RESPeCT Summer Institute</p>	<p><b>Display Slide 1.</b> RESPeCT PD Program (5 min)</p> <p>a. Take care of any housekeeping issues.</p>
		 <p><b>Agenda for Day 3</b></p> <ul style="list-style-type: none"> <li>• Day-2 reflections</li> <li>• Focus questions</li> <li>• Purposes and key features of STL strategies 4 and 5</li> <li>• Lesson analysis: STL strategies 4 and 5</li> <li>• Lunch</li> <li>• Content deepening: properties of matter</li> <li>• Summary, homework, and reflections</li> </ul>	<p><b>Display Slide 2.</b> Agenda for Day 3 (2 min)</p> <p>a. Talk through the agenda for the day.</p>

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	<p>agenda, the focus questions, and the theme for the day and the rest of the week: <i>moving student thinking forward</i>.</p> <ul style="list-style-type: none"> <li>Review Summary of STeLLA Student Thinking Lens Strategies in the STeLLA strategies booklet and recognize two patterns:               <ol style="list-style-type: none"> <li>Some strategies are designed only to reveal student thinking (strategies 1 and 2), while most are also designed to challenge student thinking.</li> <li>The Student Thinking Lens includes three questioning strategies and five activity strategies.</li> </ol> </li> </ul> <p><b>Posters and Charts</b></p> <ul style="list-style-type: none"> <li>STeLLA Framework and Strategies poster</li> <li>Day-3 Agenda (chart)</li> <li>Day-3 Focus Questions (chart)</li> <li>Strategy charts from day 1 (STL strategies 1–3)</li> </ul> <p><b>PD Resources</b></p> <ul style="list-style-type: none"> <li>STeLLA strategies booklet</li> </ul>	<div data-bbox="846 267 1312 755"> <p><b>Trends in Reflections</b></p> <table border="1"> <thead> <tr> <th data-bbox="877 337 1087 358">Lesson Analysis</th> <th data-bbox="1087 337 1281 358">Science Content Learning</th> </tr> </thead> <tbody> <tr><td> </td><td> </td></tr> <tr><td> </td><td> </td></tr> <tr><td> </td><td> </td></tr> <tr><td> </td><td> </td></tr> <tr><td> </td><td> </td></tr> <tr><td> </td><td> </td></tr> <tr><td> </td><td> </td></tr> <tr><td> </td><td> </td></tr> <tr><td> </td><td> </td></tr> </tbody> </table> </div> <div data-bbox="846 771 1312 1315"> <p><b>Today's Focus Questions</b></p> <table border="0"> <tr> <td data-bbox="877 852 1071 1031"> <p><b>Lesson Analysis</b></p> <ul style="list-style-type: none"> <li>How can analyzing data and constructing explanations help students <b>move forward</b> toward deeper understandings of science ideas?</li> </ul> </td> <td data-bbox="1102 852 1281 1096"> <p><b>Content Deepening</b></p> <ul style="list-style-type: none"> <li>Are physical changes reversible at room temperature? What is your evidence?</li> <li>Are atoms created or destroyed when matter undergoes a physical or chemical change? How can you prove your ideas?</li> </ul> </td> </tr> </table> </div>	Lesson Analysis	Science Content Learning																			<p><b>Lesson Analysis</b></p> <ul style="list-style-type: none"> <li>How can analyzing data and constructing explanations help students <b>move forward</b> toward deeper understandings of science ideas?</li> </ul>	<p><b>Content Deepening</b></p> <ul style="list-style-type: none"> <li>Are physical changes reversible at room temperature? What is your evidence?</li> <li>Are atoms created or destroyed when matter undergoes a physical or chemical change? How can you prove your ideas?</li> </ul>	<p><b>Display Slide 3.</b> Trends in Reflections (5 min)</p> <ol style="list-style-type: none"> <li>Invite participants to look at your feedback on their reflections from day 2 and offer reactions, comments, or follow-up questions.</li> <li><b>Optional:</b> Give participants an opportunity to refine the norms for working together.</li> </ol> <p><b>Display Slide 4.</b> Today's Focus Questions (2 min)</p> <ol style="list-style-type: none"> <li>Introduce the focus questions that will guide today's session.</li> <li>"The words <i>move forward</i> are in bold on the slide because that's our theme for today and the rest of the week. Yesterday we practiced asking elicit and probe questions, which are great for revealing student ideas. But what do we do with those ideas once we've elicited them? How do we support students in moving forward toward deeper understandings of science ideas?"</li> </ol>
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		 <p><b>STeLLA Conceptual Framework</b> Learning to analyze science teaching through two lenses</p> <p>allows you to learn and use strategies for more effective science teaching</p> <p><b>SCIENCE TEACHING</b></p> <table border="1"> <tr> <td> <b>STRATEGIES TO REVEAL, SUPPORT, AND CHALLENGE STUDENT THINKING</b>            1. Ask questions to elicit student ideas and predictions.            2. Ask questions to probe student ideas and predictions.            3. Ask questions to challenge student thinking.            4. Engage students in analyzing and interpreting data and observations.            5. Engage students in constructing explanations and arguments.            6. Engage students in using and applying new science ideas in a variety of ways and contexts.            7. Engage students in making connections by synthesizing and summarizing key science ideas.            8. Engage students in communicating in scientific ways.         </td> <td> <b>STRATEGIES TO CREATE A COHERENT SCIENCE CONCEPT STRUCTURE</b>            A. Identify one main learning goal.            B. Set the purpose with a focus question or goal statement.            C. Select activities that are matched to the learning goal.            D. Select content representations and models matched to the learning goal and engage students in their use.            E. Sequence key science ideas and activities appropriately.            F. Make explicit links between science ideas and activities.            G. Link science ideas to other science ideas.            H. Highlight key science ideas and focus question throughout.            I. Summarize key science ideas.         </td> </tr> </table>	<b>STRATEGIES TO REVEAL, SUPPORT, AND CHALLENGE STUDENT THINKING</b> 1. Ask questions to elicit student ideas and predictions. 2. Ask questions to probe student ideas and predictions. 3. Ask questions to challenge student thinking. 4. Engage students in analyzing and interpreting data and observations. 5. Engage students in constructing explanations and arguments. 6. Engage students in using and applying new science ideas in a variety of ways and contexts. 7. Engage students in making connections by synthesizing and summarizing key science ideas. 8. Engage students in communicating in scientific ways.	<b>STRATEGIES TO CREATE A COHERENT SCIENCE CONCEPT STRUCTURE</b> A. Identify one main learning goal. B. Set the purpose with a focus question or goal statement. C. Select activities that are matched to the learning goal. D. Select content representations and models matched to the learning goal and engage students in their use. E. Sequence key science ideas and activities appropriately. F. Make explicit links between science ideas and activities. G. Link science ideas to other science ideas. H. Highlight key science ideas and focus question throughout. I. Summarize key science ideas.	<p><b>Display Slide 5.</b> STeLLA Conceptual Framework (1 min)</p> <p>a. Point out the strategies highlighted on the slide.</p> <p>b. “We’ll continue working on understanding and using the Student Thinking Lens <i>questioning</i> strategies, but today we’ll focus on two closely related <i>activity</i> strategies. Strategy 4 engages students in analyzing and interpreting data and observations, and strategy 5 engages students in constructing explanations and arguments.”</p>
<b>STRATEGIES TO REVEAL, SUPPORT, AND CHALLENGE STUDENT THINKING</b> 1. Ask questions to elicit student ideas and predictions. 2. Ask questions to probe student ideas and predictions. 3. Ask questions to challenge student thinking. 4. Engage students in analyzing and interpreting data and observations. 5. Engage students in constructing explanations and arguments. 6. Engage students in using and applying new science ideas in a variety of ways and contexts. 7. Engage students in making connections by synthesizing and summarizing key science ideas. 8. Engage students in communicating in scientific ways.	<b>STRATEGIES TO CREATE A COHERENT SCIENCE CONCEPT STRUCTURE</b> A. Identify one main learning goal. B. Set the purpose with a focus question or goal statement. C. Select activities that are matched to the learning goal. D. Select content representations and models matched to the learning goal and engage students in their use. E. Sequence key science ideas and activities appropriately. F. Make explicit links between science ideas and activities. G. Link science ideas to other science ideas. H. Highlight key science ideas and focus question throughout. I. Summarize key science ideas.				
		<p><b>The Student Thinking Lens: Moving Student Thinking Forward</b></p> <p><i>How can we advance students’ science learning without just telling them about science ideas and expecting them to memorize the concepts?</i></p> <p>By using STeLLA strategies 4–8 to engage students in making sense of the world around them.</p> 	<p><b>Display Slide 6.</b> The Student Thinking Lens: Moving Student Thinking Forward (10 min)</p> <p>a. Initially, reveal <b>only</b> the question on the slide.</p> <p>b. Have participants think about the question for a minute; then open up a brief conversation about it.</p> <p>c. Ask the following questions to stimulate discussion if participants are struggling:</p> <ul style="list-style-type: none"> <li>• What was your experience as a science student in school or college?</li> <li>• How were you expected to learn science ideas? What learning methods were used?</li> <li>• Did you ever have the opportunity in science classes to make sense of the experiments you performed (instead of just recording the correct answers in a lab report)?</li> <li>• Did science teachers ever support your learning in ways that went beyond merely having you take lecture notes, read from a textbook, or record the correct answers in lab reports?</li> </ul>		

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PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process		
		<p><b>The Student Thinking Lens: From Questions to Activities</b></p> <ul style="list-style-type: none"> <li>• Look at the Summary of STeLLA Student Thinking Lens Strategies in the strategies booklet.</li> <li>• What distinguishes strategies 1–3 from the rest of the Student Thinking Lens strategies?</li> </ul>	<p><b>Display Slide 8.</b> The Student Thinking Lens: From Questions to Activities (5 min)</p> <p>a. <b>Individuals:</b> Have participants briefly examine the summary chart of STL strategies in the STeLLA strategies booklet (Summary of STeLLA Student Thinking Lens Strategies).</p> <p><b>Note:</b> Direct participants to the correct page in the strategies booklet or have them consult the table of contents.</p> <p>b. <b>Whole group:</b> “How are the first three strategies different from the rest?”</p> <p><b>Key ideas:</b></p> <ul style="list-style-type: none"> <li>• Strategies 1–3 are questions; the rest are activities.</li> <li>• Probe and challenge questions can and should be asked during all types of activities.</li> </ul>		
<p>8:35–9:35 60 min</p> <p><b>Introducing Student Thinking Lens (STL) Strategies 4 and 5</b></p> <p>Slides 9–11</p>	<p><b>Purpose</b></p> <ul style="list-style-type: none"> <li>• Develop an initial understanding of strategy 4: Engage students in analyzing and interpreting data and observations.</li> <li>• Develop an initial understanding of strategy 5: Engage students in constructing explanations and arguments.</li> <li>• Examine the relationships among the science practices of observing, analyzing and interpreting, and constructing explanations and arguments.</li> </ul> <p><b>Content</b></p> <ul style="list-style-type: none"> <li>• STL strategy 4 engages students</li> </ul>	<p><b>STL Strategies 4 and 5: Purposes and Key Features</b></p> <table border="1" data-bbox="871 1003 1272 1092"> <tr> <td data-bbox="871 1003 1073 1092"> <p><b>Strategy 4</b></p> <p>What are the purpose and key features?</p> </td> <td data-bbox="1073 1003 1272 1092"> <p><b>Strategy 5</b></p> <p>What are the purpose and key features?</p> </td> </tr> </table>	<p><b>Strategy 4</b></p> <p>What are the purpose and key features?</p>	<p><b>Strategy 5</b></p> <p>What are the purpose and key features?</p>	<p><b>Display Slide 9.</b> STL Strategies 4 and 5: Purposes and Key Features (30 min)</p> <p>a. <b>Small groups (12 min):</b> Divide participants into two groups and assign one strategy to each group. Have one group create a chart listing the purpose and key features of strategy 4, and have the other group chart the purpose and key features of strategy 5. Each group should be prepared to answer the discussion question for the assigned strategy.</p> <p>b. <b>Whole-group share-out (18 min):</b> Have groups report on the purpose and key features of each strategy.</p> <p><b>Key ideas:</b></p> <ul style="list-style-type: none"> <li>• Strategy 4 involves activities that engage</li> </ul>
<p><b>Strategy 4</b></p> <p>What are the purpose and key features?</p>	<p><b>Strategy 5</b></p> <p>What are the purpose and key features?</p>				

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
	<p>in analyzing and interpreting data and observations. Activities involve organizing data and/or observations, identifying patterns, and looking for meaning in the data.</p> <ul style="list-style-type: none"> <li>STL strategy 5 engages students in constructing explanations and arguments. Activities involve using logical thinking, evidence, and science ideas to construct explanations of scientific data or observed phenomena, as well as critiquing proposed explanations using scientific argumentation.</li> </ul> <p><b>What Participants Do</b></p> <ul style="list-style-type: none"> <li>Create and discuss strategy charts summarizing the purposes and key features of strategies 4 and 5.</li> <li>Discuss the differences and relationships among observing, analyzing and interpreting, and constructing explanations and arguments.</li> <li>Use written scenarios to practice identifying instances of observing, analyzing and interpreting, and constructing explanations and arguments.</li> </ul> <p><b>Handouts in PD Binder</b></p> <ul style="list-style-type: none"> <li>3.1 Quick Reference Tools for Strategies 4 and 5</li> <li>3.2 Practice Identifying Strategies 4 and 5</li> </ul>	<p><b>Relationships between Strategies 4 and 5</b></p> <p>Discuss the question assigned to your group and be ready to share your ideas:</p> <p><b>Group 1:</b> How is analyzing/interpreting different from describing observations?</p> <p><b>Group 2:</b> How are strategy 4 and strategy 5 different? How are they related?</p> <p><b>Group 3:</b> How are scientific explanation and scientific argumentation related? How are they different? How are arguments in science different from arguments in everyday situations?</p> <p><small>To support your responses, use the STeLLA strategies booklet and Quick Reference Tools for Strategies 4 and 5 (handout 3.1).</small></p>	<p>students in organizing their data and/or observations and looking for patterns and meaning in them. They aren't just "doing" activities or describing their observations.</p> <ul style="list-style-type: none"> <li>Strategy 5 engages students in learning how to use logical thinking, evidence, and science ideas to construct explanations of scientific data or phenomena they have observed. It also engages them in critiquing various proposed explanations through scientific argumentation.</li> <li>Remind participants that these strategies are closely related and will overlap in some activities. However, each has a specific purpose and unique attributes.</li> </ul> <p><b>Display Slide 10.</b> Relationships between Strategies 4 and 5 (15 min)</p> <p>a. <b>Small groups (5 min):</b> Divide participants into three small groups or pairs. Assign each group one question to discuss and tell participants to be ready to share their ideas with the entire group.</p> <p>b. <b>Emphasize:</b> Participants should use the STeLLA strategies booklet and Quick Reference Tools for Strategies 4 and 5 (PD handout 3.1) to support their responses.</p> <p>c. <b>Whole-group share-out (10 min):</b></p> <ul style="list-style-type: none"> <li>"What did you come up with for the first question?"</li> </ul> <p><b>Key ideas for question 1:</b> Analysis and interpretation involve moving beyond simply describing observations to <i>doing</i> something with the data, including (but not limited to) making comparisons, identifying relationships, and organizing data in ways that will reveal patterns</p>

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	<p><b>PD Leader Masters</b></p> <ul style="list-style-type: none"> <li>• PD Leader Master: Practice Identifying Strategies 4 and 5</li> </ul> <p><b>Supplies</b></p> <ul style="list-style-type: none"> <li>• Chart paper and markers</li> </ul> <p><b>PD Resources</b></p> <ul style="list-style-type: none"> <li>• STeLLA strategies booklet</li> </ul>		<p>(such as using charts, diagrams, and graphs).</p> <ul style="list-style-type: none"> <li>• “What did you come up with for the second question?”</li> </ul> <p><b>Key ideas for question 2:</b> Strategy 4 lays the groundwork for strategy 5. Before we can build a scientific explanation for a specific phenomenon, we need to make some observations, analyze the data to reveal patterns, and organize the data to gather the necessary evidence to support construction of a scientific explanation. A scientific explanation includes a claim that answers the question being studied, evidence that supports the claim, and reasoning that links the claim to the evidence and to science ideas.</p> <ul style="list-style-type: none"> <li>• “What did you come up with for the third question?”</li> </ul> <p><b>Key ideas for question 3:</b> A scientific explanation includes a claim that answers the question being studied, evidence that supports the claim, and reasoning that links the claim to the evidence and to science ideas. Scientific arguments involve assessing the strength and quality of the evidence and reasoning in different scientific explanations for the same observations and determining which proposed explanation has the best supporting evidence, science ideas, and reasoning.</p>

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		<p><b>Practice Identifying Strategies 4 and 5</b></p> <p>Examine student statements made during a science-class activity. Decide whether each statement represents the following:</p> <ul style="list-style-type: none"> <li>• An observation</li> <li>• An analysis or interpretation of the observations (e.g., describing a pattern) (strategy 4)</li> <li>• An attempt to construct an explanation that has a claim, some evidence, and/or reasoning that uses science ideas (strategy 5)</li> <li>• An attempt to construct an argument (strategy 5)</li> </ul> <p style="text-align: right;"><small>Refer to Practice Identifying Strategies 4 and 5 (handout 3.2).</small></p>	<p><b>Display Slide 11.</b> Practice Identifying Strategies 4 and 5 (15 min)</p> <p>a. “Before we view classroom video clips to identify and analyze strategies 4 and 5, we’re going to practice identifying observations, analyses, interpretations, explanations, and arguments from a handout of student statements. Learning to distinguish which strategy students are using in these examples will help us when we review the classroom videos, where the strategies aren’t always as clear cut.”</p> <p>b. Refer participants to handout 3.2 in their PD program binders (Practice Identifying Strategies 4 and 5).</p> <p>c. <b>Pairs:</b> Have participants work in pairs to analyze student statements in the handout.</p> <p>d. <b>Whole group:</b> As participants discuss and clarify their analyses of the student statements, encourage them to refer frequently to the STeLLA strategies booklet and the Quick Reference Tools handout (PD handout 3.1).</p> <p><b>Note:</b> For examples of ideal participant responses, see PD Leader Master: Practice Identifying Strategies 4 and 5.</p>

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
9:35–12:00 145 min (Includes 10-min break)  <b>Lesson            Analysis:            STL            Strategies            4 and 5</b>  Slides 12–22	<b>Purpose</b> <ul style="list-style-type: none"> <li>Use lesson analysis of classroom videos to better understand strategies 4 and 5, how they're related, and how they can challenge student thinking to move forward.</li> <li>Deepen science-content knowledge of properties of matter through lesson analysis.</li> </ul> <b>Content</b> <ul style="list-style-type: none"> <li>STL strategy 4 engages students in analyzing and interpreting data and observations. Activities involve organizing data and/or observations, identifying patterns, and looking for meaning in the data.</li> <li>STL strategy 5 engages students in constructing explanations and arguments. Activities involve using logical thinking, evidence, and science ideas to construct explanations of scientific data or observed phenomena, as well as critiquing proposed explanations using scientific argumentation.</li> </ul> <b>What Participants Do</b> <ul style="list-style-type: none"> <li>Watch one classroom video clip to identify strategy 4 and analyze student thinking that this strategy reveals and challenges.</li> <li>Examine transcript excerpts in the STeLLA strategies booklet for practice identifying strategies 4 and 5.</li> </ul>	<div style="background-color: #e0e0e0; padding: 5px; margin-bottom: 10px;"></div> <p><b>Lesson Analysis Focus Question</b></p> <p>How can analyzing data and constructing explanations help students <i>move forward</i> toward deeper understandings of science ideas?</p>	<p><b>Display Slide 12.</b> Lesson Analysis Focus Question (Less than 1 min)</p> <p>a. Review the focus question that will guide today's lesson analysis work.</p>
	<ul style="list-style-type: none"> <li>STL strategy 4 engages students in analyzing and interpreting data and observations. Activities involve organizing data and/or observations, identifying patterns, and looking for meaning in the data.</li> <li>STL strategy 5 engages students in constructing explanations and arguments. Activities involve using logical thinking, evidence, and science ideas to construct explanations of scientific data or observed phenomena, as well as critiquing proposed explanations using scientific argumentation.</li> </ul>	<div style="background-color: #e0e0e0; padding: 5px; margin-bottom: 10px;"></div> <p><b>Lesson Analysis: Review Lesson Context</b> <span style="float: right; font-size: small;">Video Clip 1</span></p> <p>Review the lesson context at the top of the transcript for video clip 1 (handout 3.3 in your PD program binder).</p>	<p><b>Display Slide 13.</b> Lesson Analysis: <b>Review</b> Lesson Context, Video Clip 1 (2 min)</p> <p>a. "Now let's see if we can recognize students analyzing and interpreting data in a classroom video clip."</p> <p>b. Review the lesson context at the top of the transcript for video clip 1 (handout 3.3 in PD binder), making sure participants understand both the content and activity in focus.</p>
	<ul style="list-style-type: none"> <li>Watch one classroom video clip to identify strategy 4 and analyze student thinking that this strategy reveals and challenges.</li> <li>Examine transcript excerpts in the STeLLA strategies booklet for practice identifying strategies 4 and 5.</li> </ul>	<div style="background-color: #e0e0e0; padding: 5px; margin-bottom: 10px;"></div> <p><b>Lesson Analysis: Identify Strategy 4</b> <span style="float: right; font-size: small;">Video Clip 1</span></p> <p><b>Identify</b> instances where the teacher or the students are engaged in <b>analyzing and interpreting data and observations</b> by</p> <ul style="list-style-type: none"> <li>clarifying key observations,</li> <li>identifying a pattern in the observations,</li> <li>identifying what needs to be explained,</li> <li>organizing data/observations, and/or</li> <li>trying to make sense of the observations (analyzing, interpreting).</li> </ul> <p><b>Discuss:</b> How are these actions implemented in the video?</p> <p style="font-size: x-small;"><a href="#">Link to video clip 1: 3.1_msocp_gr2_matter_griffin_L2_c4-c6</a></p>	<p><b>Display Slide 14.</b> Lesson Analysis: <b>Identify</b> Strategy 4, Video Clip 1 (25 min)</p> <p>a. "As we watch the video clip, we'll <b>identify</b> actions that illustrate strategy 4. Be on the lookout for instances where the teacher or the students do something listed on the slide. That's what we'll discuss first."</p> <p>b. Show the video clip.</p> <p>c. <b>Individuals:</b> "Think about the strategy 4 actions</p>

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
	<ul style="list-style-type: none"> <li>• Watch a second classroom video clip to identify strategy 5 and analyze student thinking that this strategy reveals and challenges.</li> <li>• Summarize key ideas about the relationships between strategies 4 and 5.</li> </ul> <p><b>Videos</b></p> <ul style="list-style-type: none"> <li>• Video Clip 3.1, Griffin classroom</li> <li>• Video Clip 3.2, Fowler classroom</li> </ul> <p><b>Handouts in PD Binder</b></p> <ul style="list-style-type: none"> <li>• 3.1 Quick Reference Tools for Strategies 4 and 5</li> <li>• 3.3 Transcript for Video Clip 3.1</li> <li>• 3.4 Transcript for Video Clip 3.2</li> </ul> <p><b>PD Resources</b></p> <ul style="list-style-type: none"> <li>• STeLLA strategies booklet</li> </ul>		<p>listed on the slide.”</p> <p>d. <b>Whole group:</b> “Discuss the question on the slide. Make sure to support your claims with evidence from the video transcript. Think about whether the teacher guided student thinking toward more-scientific understandings.”</p> <p><b>Observations:</b></p> <ul style="list-style-type: none"> <li>• <b>Clarifying key observations:</b> video segments 00:00:34; 00:00:36; 00:00:57; 00:01:35–01:40; 00:02:50</li> <li>• <b>Identifying a pattern in observations:</b> segments 00:01:02; 00:02:41–02:44; 00:02:57</li> <li>• <b>Identifying what needs to be explained:</b> segment 00:03:03</li> <li>• <b>Organizing observations:</b> segment 00:02:44</li> <li>• <b>Interpreting data:</b> segments 00:00:38–00:46; 00:01:14–01:28; 00:01:55–02:29; 00:03:05–03:36</li> </ul> <p><b>Ideal responses:</b></p> <ul style="list-style-type: none"> <li>• At video segment 00:00:38, the teacher asks a question (“Why is it turning back into a solid?”) that challenges students to connect their observations of the three materials to the idea that heat has to be removed—the substance must be cooled—for a liquid to turn back into a solid.</li> <li>• At segment 00:01:55, the teacher uses another challenge question (“Who has an idea of whether or not the ice melted at a higher temperature or a lower temperature?”) to guide students toward the idea that different substances have different melting points. However, the connection wasn’t made in this clip.</li> </ul> <p><b>Content note:</b> Ice melts at 32 degrees Fahrenheit, so it wasn’t necessary to turn the hot plate up to 300 degrees. The teacher’s action confused some students, who thought the ice melted at a higher temperature than the other substances instead of a</p>

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
		<p data-bbox="871 342 1203 391">Lesson Analysis: <b>Analyze Strategy 4 and Reflect</b></p> <p data-bbox="1234 350 1283 378">Video Clip 1</p> <p data-bbox="871 407 932 423"><b>Analyze</b></p> <ul data-bbox="892 431 1274 509" style="list-style-type: none"> <li>• What student thinking is revealed in the video clip by engaging students in analysis and interpretation?</li> <li>• Were any opportunities missed for engaging students in analyzing and interpreting data and observations?</li> </ul> <p data-bbox="871 518 932 534"><b>Reflect</b></p> <ul data-bbox="892 542 1274 643" style="list-style-type: none"> <li>• What did you learn about strategy 4 from analyzing this video clip?</li> <li>• Did the analysis process focus your attention on aspects you might not have noticed before? If yes, what is one example?</li> </ul>	<p data-bbox="1335 256 1560 280">lower temperature.</p> <p data-bbox="1335 321 1871 378"><b>Display Slide 15.</b> Lesson Analysis: <b>Analyze Strategy 4 and Reflect</b>, Video Clip 1 (25 min)</p> <p data-bbox="1335 431 1948 638">a. <b>Individuals:</b> “For the first analysis question on the slide, study the transcript for video clip 1 and come up with a claim, evidence, and reasoning to support your claim. For the second analysis question, consider alternative moves the teacher could have made as you identify missed opportunities.”</p> <p data-bbox="1335 659 1929 776">b. <b>Whole group:</b> After participants have shared their analyses, ask, “Were there any missed opportunities for engaging students in analyzing and interpreting data?”</p> <p data-bbox="1335 797 1934 881">c. <b>Reflect:</b> Discuss the reflection questions on the slide, making sure participants share specifically what they learned about strategy 4.</p> <p data-bbox="1335 902 1514 927"><b>Observations:</b></p> <ul data-bbox="1335 935 1944 1446" style="list-style-type: none"> <li>• <b>Possible claim:</b> Matter changes to a liquid, or melts, when it’s heated and changes back to a solid when heat is removed. <i>[Most students understand that the three substances melt when they’re heated. They also observe that the first two substances are beginning to change back to solids after the heat is removed. ]</i></li> <li>• <b>Evidence:</b> At segment 00:02:41, a student observes that all three substances melt and freeze. Other students observe that matter gets stickier as it changes back to a solid (Michaela at video segments 00:00:05–00:15; Rowan at segment 00:01:14). And at segments 00:00:44 and 00:01:14, students connect cooling with the crayon wax changing from a liquid to a solid.</li> <li>• <b>Missed opportunities:</b> <ul data-bbox="1381 1422 1854 1446" style="list-style-type: none"> <li>• The teacher missed an opportunity to</li> </ul> </li> </ul>

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
			<p>organize the temperature data on the data table and engage students in making sense of the patterns in the data. At segment 00:01:55, before asking the question “So who has an idea of whether or not the ice melted at a higher temperature or a lower temperature?” the teacher could have entered the temperature of the melting ice on the data table. This would have involved clarifying the meaning of <i>melting point</i>—the temperature of the water when the ice melted—in contrast to the hot-plate temperature (300 degrees).</p> <ul style="list-style-type: none"> <li>• There are also missed opportunities to probe student thinking. At segment 00:02:22, the student correctly answered that the melting temperature of ice is lower than 300 degrees, but her reasoning isn’t clear. A probe question could have prompted her to elaborate. The student at segment 00:03:05 has a good explanation for why the melting point of water is 63 degrees in this experiment, but this idea wasn’t explored further.</li> <li>• The teacher missed another opportunity to emphasize the pattern that cooling is involved in changing liquids to solids, and heating is involved in changing solids to liquids. After one student observed that all three substances freeze and melt (segment 00:02:41), the teacher could have linked this comment to the temperature data and asked a question that would have challenged the student to analyze the data: “Yes, all three substances melt and freeze, but do they melt and freeze in the same way? What does our data table tell us?”</li> </ul>



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		<p><b>Strategy 5 Practice: Explanation and Argumentation</b></p> <p>Analyze the sample transcript in the strategies booklet to find evidence of students engaged in <b>constructing explanations and arguments</b> by</p> <ul style="list-style-type: none"> <li>• making a claim that answers the investigation question,</li> <li>• making a claim and supporting it with evidence,</li> <li>• making a claim and supporting it with science ideas,</li> <li>• using logical reasoning to explain why the evidence supports a claim, and/or</li> <li>• making an argument.</li> </ul>	<p><b>Display Slide 16.</b> Strategy 5 Practice: Explanation and Argumentation (10 min)</p> <p>a. “Strategy 5 is the focus of the next video clip, although you may also see evidence of strategy 4 being used.”</p> <p>b. Have participants analyze the second transcript example (under “About Weather” in the strategy 5 chapter) in the STeLLA strategies booklet and look for evidence of students engaging in constructing explanations and arguments.</p> <p><b>Note:</b> This is an important activity, but it can be cut if time is short. The sample dialogue comes from a lesson on weather for 1st graders or kindergarteners. Keep in mind that in an actual classroom conversation, most students wouldn’t provide such complete explanations or arguments, so the teacher would need to use questioning strategies and sentence starters to help students develop stronger explanations and arguments.</p> <p>c. “Before we view another classroom video, let’s practice analyzing an example of strategy 5 in the STeLLA strategies booklet. Read the sample transcript in the ‘About Weather’ section and see if you can find any evidence of the teacher engaging students in constructing explanations and arguments. Refer to the action list on the slide for guidance.”</p> <p>d. <b>Individual work time (5 min).</b></p> <p>e. <b>Whole-group share-out:</b> Have participants share evidence from the transcript of students engaging in strategy 5, noting the specific action illustrated from the list on the slide.</p> <p><b>Ideal responses:</b></p>

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			<ul style="list-style-type: none"> <li>• Students in the sample transcript aren't yet using science ideas to support their claims. This will come later in the lesson sequence. In the following examples, students (a) make a claim, (b) provide evidence to support the claim, (c) use logical reasoning to support the claim, or (d) engage in argumentation. <ul style="list-style-type: none"> <li>• <b>Student 1:</b> <ul style="list-style-type: none"> <li>• <i>Claim:</i> "I think our place is sunnier than Place B."</li> <li>• <i>Evidence:</i> "Our place has more sunny days than cloudy days.... There were 18 sunny days this month and only five cloudy days."</li> </ul> </li> <li>• <b>Student 2:</b> <ul style="list-style-type: none"> <li>• <i>Claim:</i> "I think that our weather is sunny during November.... Place B isn't sunny at all."</li> <li>• <i>Evidence:</i> "Because 18 is more than five, so that means it's sunny. [In Place B,] almost all their days are cloudy."</li> </ul> </li> <li>• <b>Student 3:</b> <ul style="list-style-type: none"> <li>• <i>Claim:</i> "I think that the weather isn't the same everywhere."</li> <li>• <i>Evidence:</i> (1) "We found out about the weather differences in these two places." (2) "We have more sunny days than Place B."</li> <li>• <i>Logical reasoning:</i> "There must be a difference between Pomona and Place B that causes us to have more sunny days than Place B. Maybe Place B is at a higher altitude, because we learned that it's cooler at higher altitudes. Maybe it's cooler because it's cloudy."</li> </ul> </li> <li>• <b>Student 4:</b> <ul style="list-style-type: none"> <li>• <i>Argument:</i> "I agree with the idea that maybe Place B is at a higher altitude."</li> <li>• <i>Evidence:</i> "I sometimes see clouds"</li> </ul> </li> </ul> </li> </ul>





PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
			<p>covering up Mount Baldy when it's sunny down here."</p> <ul style="list-style-type: none"> <li>• <b>Student 5:</b> <ul style="list-style-type: none"> <li>• <i>Argument:</i> "Place B could be cloudy for another reason."</li> <li>• <i>Logical reasoning:</i> "Maybe it's more polluted in Place B. Pollution causes smog."</li> </ul> </li> <li>• <b>Student 6:</b> <ul style="list-style-type: none"> <li>• <i>Argument and evidence:</i> "But it's really polluted here, and we have lots of sunny days."</li> </ul> </li> <li>• <b>Student 2:</b> <ul style="list-style-type: none"> <li>• <i>Argument:</i> "I agree with S3 that weather isn't the same everywhere, and I agree with his evidence, but I have a different reason."</li> <li>• <i>Evidence:</i> "We went to San Francisco, and it was, like, cloudy and foggy every morning."</li> <li>• <i>Logical reasoning:</i> "I think it was because it was right next to the ocean."</li> </ul> </li> </ul>
<b>10-MINUTE BREAK</b>			
		<div style="border: 1px solid gray; padding: 5px;"> <p><b>Lesson Analysis: Review Lesson Context</b> <span style="float: right; font-size: small;">Video Clip 2</span></p> <p>Review the lesson context at the top of the transcript for video clip 2 (handout 3.4 in your PD program binder).</p> </div>	<p><b>Display Slide 17.</b> Lesson Analysis: <b>Review</b> Lesson Context, Video Clip 2 (1 min)</p> <ol style="list-style-type: none"> <li>a. "Now we're going to look at another video clip and focus on identifying strategy 5: Engage students in constructing explanations and arguments."</li> <li>b. Read the context of the lesson at the top of the transcript for video clip 2 (handout 3.4 in the PD program binder).</li> </ol>

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
		<p style="text-align: right; margin-right: 20px;"><small>Video Clip 2</small></p> <p><b>Lesson Analysis: Identify Strategy 5</b></p> <p><b>Identify</b> instances in the video clip where students are <b>constructing explanations or arguments</b> by</p> <ul style="list-style-type: none"> <li>• stating an explanation or claim,</li> <li>• using evidence from observations to support or develop the explanation/claim,</li> <li>• using science ideas to support or develop the explanation/claim,</li> <li>• using logical reasoning to develop the explanation/claim, and/or</li> <li>• engaging in argumentation (agreeing, disagreeing).</li> </ul> <p><b>Discuss:</b> How are these actions implemented in the video?</p> <p style="text-align: center;"><a href="#">Link to video clip 2: 3.2_mspcp_gr2_matter_fowler_LS_c7-c9</a></p>	<p><b>Display Slide 18.</b> Lesson Analysis: <b>Identify</b> Strategy 5, Video Clip 2 (25 min)</p> <p>a. “As you watch the video clip, <b>identify</b> instances where students are engaged in constructing explanations and arguments (strategy 5). You might notice examples of strategy 4 (analyzing and interpreting data), but focus on identifying strategy 5. Also notice the kinds of questions the teacher asks (elicit, probe, or challenge).”</p> <p>b. Before showing the video clip, read the list of actions on the slide.</p> <p>c. <b>Individuals:</b> “Think about the strategy 5 actions listed on the slide.”</p> <p>d. <b>Whole group:</b> “Discuss the question on the slide. Make sure to support your claims with evidence from the video transcript.”</p> <p>e. <b>Emphasize:</b> “Strategy 5 is designed to help move student thinking forward toward deeper understandings of science ideas, so we should see challenge questions as well as probe questions in the video clip.”</p> <p><b>Observations:</b></p> <ul style="list-style-type: none"> <li>• At video segments 00:03:19–03:29, Ashlynn makes a claim that matter (atoms and molecules) isn’t created or destroyed when it changes from a liquid to a solid but stays the same.</li> <li>• Ashlynn supports her claim using observational evidence: “The Legos in the ice and the water were the same” (segment 00:03:34); the Lego ice and water molecules were balanced on the scale (segment 00:03:42); and the Lego ice and water molecules were the same weight (00:03:46).</li> <li>• At segments 00:03:19–04:03, Ashlynn uses science ideas to support her claim. She doesn’t</li> </ul>

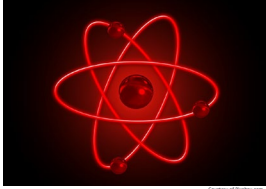
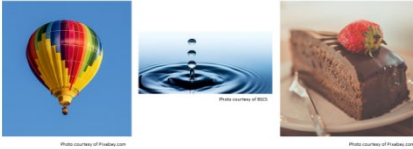
PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
		<p data-bbox="873 613 1184 667">Lesson Analysis: <b>Analyze</b> Strategy 5 and <b>Reflect</b></p> <p data-bbox="1230 623 1276 651">Video Clip 2</p> <p data-bbox="873 678 930 695"><b>Analyze</b></p> <ul data-bbox="894 701 1289 781" style="list-style-type: none"> <li>• What student thinking is revealed by engaging students in constructing explanations of the conservation of matter?</li> <li>• Were there any missed opportunities to support students in constructing explanations and arguments?</li> </ul> <p data-bbox="873 787 930 803"><b>Reflect</b></p> <ul data-bbox="894 810 1276 906" style="list-style-type: none"> <li>• What did you learn about strategy 5 from analyzing this video clip?</li> <li>• Did the analysis process focus your attention on aspects you might not have noticed before? If yes, what is one example?</li> </ul>	<p data-bbox="1362 256 1948 375">explicitly mention atoms and molecules, but when she responds to the teacher’s question at 00:03:19, we can assume that’s what she’s referring to.</p> <ul data-bbox="1335 381 1944 558" style="list-style-type: none"> <li>• At segment 00:03:59, Ashlynn uses logical reasoning to develop her explanation by showing that if atoms or molecules had been created or destroyed, the balance scale would have tipped, indicating that the Lego ice and water molecules had different weights.</li> </ul> <p data-bbox="1335 594 1871 651"><b>Display Slide 19.</b> Lesson Analysis: <b>Analyze</b> Strategy 5 and <b>Reflect</b>, Video Clip 2 (25 min)</p> <p data-bbox="1335 704 1938 911">a. <b>Individuals:</b> “For the first analysis question on the slide, study the video transcript and come up with a claim, evidence, and reasoning to support your claim. For the second analysis question, consider alternative moves the teacher could have made as you identify any missed opportunities.”</p> <p data-bbox="1335 932 1898 1050">b. <b>Whole group:</b> After participants have shared their analyses, ask, “Were there any missed opportunities for engaging students in constructing explanations and arguments?”</p> <p data-bbox="1335 1071 1934 1157">c. <b>Reflect:</b> Discuss the reflection questions on the slide, making sure participants share specifically what they learned about strategy 5.</p> <p data-bbox="1335 1179 1514 1203"><b>Observations:</b></p> <ul data-bbox="1335 1209 1948 1446" style="list-style-type: none"> <li>• <b>Possible claim:</b> Matter isn’t created or destroyed when it changes from a solid to a liquid or from a liquid to a solid. <i>[Students understand that the state of matter (whether the water is frozen or at room temperature) doesn’t contribute to either bottles’ weight, which confirms the idea that matter is neither created nor destroyed.]</i></li> <li>• <b>Evidence:</b> Students seem to agree that both the</li> </ul>


PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
			<p>ice and the water weigh the same, as well as the Lego ice and water molecules (video segments 00:00:18–00:00:26; 00:00:47; 00:02:08; 00:02:38; 00:02:40). At segment 00:02:06, Jilissa notes that no matter how you arrange the Legos on the balance, they still balance (weigh the same). It isn't clear whether students understand that the Legos represent atoms and molecules. The teacher is the only person who actually uses the words <i>atoms</i> and <i>molecules</i> in this clip (e.g., segments 00:02:44; 00:03:19). Ashlynn seems to understand how their evidence supports the idea that matter isn't gained or lost when ice changes to water (segments 00:03:19–03:59).</p> <ul style="list-style-type: none"> <li>• <b>Missed opportunities to construct explanations or arguments:</b> <ul style="list-style-type: none"> <li>• At video segment 00:00:31, the teacher could have challenged students to explain their observations about water and ice by asking, "So was any matter lost or gained when the water in the bottle froze and changed to ice? Make a claim and support it with evidence."</li> <li>• When one group says that the number of atoms is different, and another group says it's the same (segments 00:01:35–01:36), the teacher could have used this opportunity to present an alternative explanation for students to consider.</li> <li>• At segments 00:02:16–02:22, after Jilissa shares her idea that the Legos weigh the same no matter how you arrange them on the balance scale, the teacher could have challenged students to disagree instead of just asking, "You guys agree?" For example, the teacher could have challenged students to construct an argument by asking, "Can anyone think of a way to rearrange the Legos so that they don't weigh the same? What</li> </ul> </li> </ul>


PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
		<p data-bbox="846 318 1314 342"><b>Reflect: Key Ideas about Lesson Analysis</b></p> <ul data-bbox="877 383 1283 643" style="list-style-type: none"> <li>• Lesson analysis slows down classroom events so we can focus on specific student thinking.</li> <li>• Making a claim based on evidence challenges us to listen carefully to what students are saying and understanding. When we make quick assessments, we might think they understand things they're actually still struggling with.</li> <li>• Even though events happen fast in classroom teaching, <b>we can get better at listening to students and making on-the-spot assessments of their understandings and confusion!</b></li> </ul> <p data-bbox="846 862 1314 886"><b>Summarizing Strategies 4 and 5</b></p> <p data-bbox="877 902 1283 1008">Create a word picture (a concept map, a thinking map, or other visual) to show how analysis and interpretation (strategy 4) are related to explanation and argumentation (strategy 5). Label any connecting arrows. Suggested words to use:</p> <ul data-bbox="894 1016 1272 1146" style="list-style-type: none"> <li>• Analyze and interpret</li> <li>• Organize</li> <li>• Argument</li> <li>• Observe/observations</li> <li>• Data</li> <li>• Patterns</li> <li>• Evidence</li> <li>• Reasoning</li> <li>• Explanation</li> <li>• Science ideas</li> <li>• Logical thinking</li> </ul>	<p data-bbox="1409 256 1955 280">does this tell us about atoms and molecules?"</p> <p data-bbox="1335 318 1944 375"><b>Display Slide 20.</b> Reflect: Key Ideas about Lesson Analysis (2 min)</p> <ol data-bbox="1335 431 1944 699" style="list-style-type: none"> <li>“Let’s reflect on some key ideas you can take away from your lesson analysis experiences. These ideas may not reflect your personal experiences with lesson analysis so far, but hopefully you’ll see their value in the lesson analysis process over time.”</li> <li>Read the key ideas on the slide.</li> <li>Ask participants for their reactions to these ideas.</li> </ol> <p data-bbox="1335 829 1944 886"><b>Display Slide 21.</b> Summarizing Strategies 4 and 5 (15 min)</p> <p data-bbox="1335 935 1776 959"><b>Note:</b> Skip this activity if time is short.</p> <ol data-bbox="1335 992 1944 1292" style="list-style-type: none"> <li><b>Individuals:</b> To summarize strategies 4 and 5, have participants work independently to create visuals that show how analysis and interpretation (strategy 4) are related to explanation and argumentation (strategy 5).</li> <li><b>Pairs:</b> “Share and compare your visuals with a partner.”</li> <li><b>Whole group:</b> “What questions did this activity raise for you?”</li> </ol>


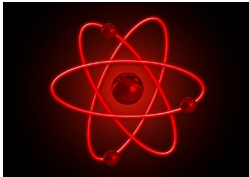
PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
		<p style="text-align: center;"><b>Reflect: Lesson Analysis Focus Question</b></p> <p>How can analyzing data and constructing explanations help students <b>move forward</b> toward deeper understandings of science ideas?</p>	<p><b>Display Slide 22.</b> Reflect: Lesson Analysis Focus Question (5 min)</p> <p>a. Review today's lesson analysis focus question.</p> <p>b. <b>Think-Pair-Share:</b> "Think for a moment about this focus question and how you might convince parents or colleagues that analyzing data and constructing explanations moves student thinking forward toward deeper understandings of science ideas. Then share your ideas with an elbow partner."</p>
12:00–12:45 45 min	<b>LUNCH</b>		
12:45–3:15 150 min (Includes 10-min break)  <b>Content Deepening: Properties of Matter</b>	<p><b>Purpose</b></p> <ul style="list-style-type: none"> <li>• Deepen participants' understandings of what matter is made of and how it can change.</li> <li>• Deepen participants' science-content knowledge by conducting investigations from the Properties of Matter lessons.</li> </ul> <p><b>Content</b></p> <ul style="list-style-type: none"> <li>• Studying the structure of water and how molecules are arranged and move in various states of</li> </ul>	<p style="text-align: center;"><b>PROPERTIES OF MATTER</b></p> <p style="text-align: center;">SCIENCE CONTENT DEEPENING <span style="float: right;">Grade 2</span></p> <div style="display: flex; justify-content: space-around; align-items: center;">     </div>	<p><b>Display Slide 23.</b> Content Deepening: Properties of Matter (Less than 1 min)</p> <p>a. "Next, we'll focus on deepening our science-content understandings of matter."</p> <p><b>Note:</b> Throughout this content deepening phase, refer as needed to the Properties of Matter Content Background Document and Common Student Ideas about Properties of Matter.</p>







PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
Slides 23–41	<p>matter and in chemical changes is essential to understanding matter.</p> <p><b>What Participants Do</b></p> <ul style="list-style-type: none"> <li>Review key science ideas about matter.</li> <li>Explore physical changes to various kinds of matter when heat is added and removed.</li> <li>Review what happens to matter and molecules during physical and chemical changes.</li> <li>Design an experiment that will help them prove whether atoms are created or destroyed when matter undergoes physical and chemical changes.</li> <li>Conduct the experiment they designed to test their ideas about conservation of matter.</li> <li>Review and discuss NGSS standards.</li> <li>Revise their answers to the unit central questions based on today’s content deepening work.</li> </ul> <p><b>Supplies</b></p> <ul style="list-style-type: none"> <li>Science notebooks</li> <li>Chart paper and markers</li> <li>Lesson materials kits</li> <li>For melting/freezing demonstration: <ul style="list-style-type: none"> <li>Small griddle</li> <li>3 clear-glass, heat-resistant beakers (Pyrex works best)</li> <li>3 wooden craft sticks (for stirring)</li> </ul> </li> </ul>	<p><b>Unit Central Questions</b></p> <p>What is matter made of? How can matter change?</p>  <hr/> <p><b>Review: What Is Matter?</b></p> <p><b>Matter</b> is anything that has mass and takes up space (or has volume). Matter exists as a liquid, a solid, or a gas. The arrangement and movement of molecules is different in each phase or state of matter.</p> 	<p><b>Display Slide 24.</b> Unit Central Questions (6 min)</p> <ol style="list-style-type: none"> <li>Review the unit central questions on the slide and ask participants to locate the initial answers they wrote in their science notebooks last time.</li> <li><b>Individuals (3 min):</b> “Take a moment to review your initial ideas for answering these questions. Then revise your answers in your science notebooks based on what we learned about matter in our previous content deepening session.”</li> <li><b>Whole group (3 min):</b> Invite participants to share their answers and ideas. Record key ideas on chart paper and ask probe and challenge questions related to matter, atoms, and molecules.</li> </ol> <hr/> <p><b>Display Slide 25.</b> Review: What Is Matter? (3 min)</p> <ol style="list-style-type: none"> <li>Review the key ideas about matter on the slide.</li> <li>“In previous sessions, we explored water as an example of matter. We know that a water molecule is made up of two hydrogen atoms and one oxygen atom, and that water molecules are arranged and move differently as a solid and a liquid. In a liquid, for instance, the molecules are loosely attracted to one another and move around more freely, flowing or sliding past each other. How would you describe the arrangement and movement of molecules in a solid like ice?”</li> </ol> <p><b>Ideal response:</b> In a solid, the molecules are tightly arranged in a rigid structure and vibrate in place. This is how a solid retains its shape and form.</p>

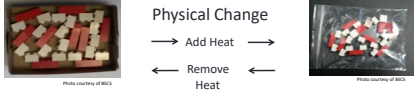
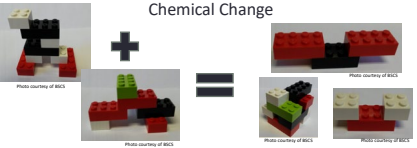
PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
	<ul style="list-style-type: none"> <li>• Hot pads or tongs</li> <li>• Chocolate chips or pieces (milk chocolate)</li> <li>• Crayons or crayon pieces (one color)</li> <li>• Butter</li> <li>• For physical-changes experiment:               <ul style="list-style-type: none"> <li>• 2 balance scales (1 per team)</li> <li>• 24 white (hydrogen) Legos and 12 red (oxygen) Legos (for Team 1)</li> <li>• 1 bottle of frozen water and 1 bottle of room-temperature water (for Team 2)</li> </ul> </li> <li>• For chemical-changes experiment:               <ul style="list-style-type: none"> <li>• 2 balance scales (1 per team)</li> <li>• 1 lime-green (sodium) Lego, 5 white (hydrogen) Legos, 5 red (oxygen) Legos, 3 black (carbon) Legos (1 set per team)</li> <li>• Bag containing baking soda and vinegar (1 per team):                   <ul style="list-style-type: none"> <li>• 1 quart-sized, sealable, plastic freezer bag</li> <li>• 1 teaspoon of baking soda</li> <li>• 2 tablespoons of white vinegar</li> <li>• Clear plastic vial with snap-on cap</li> <li>• Paper towels</li> </ul> </li> </ul> </li> </ul> <p><b>PD Resources</b></p>	<div style="border: 1px solid gray; padding: 5px; margin-bottom: 10px;"> <p style="text-align: center;"><b>Content Deepening Focus Questions</b></p> <ol style="list-style-type: none"> <li>1. Are physical changes reversible at room temperature? What is your evidence?</li> <li>2. Are atoms created or destroyed when matter undergoes a physical or chemical change? How can you prove your ideas?</li> </ol> </div> <div style="border: 1px solid gray; padding: 5px;"> <p style="text-align: center;"><b>Investigating Physical Changes in Matter</b></p> <p>Watch closely as these substances melt and record your observations in your science notebooks.</p> <div style="display: flex; justify-content: space-around; align-items: center;">  </div> <p>Keep our first focus question in mind: <i>Are physical changes reversible at room temperature? What is your evidence?</i></p> </div>	<p><b>Display Slide 26.</b> Content Deepening Focus Questions (Less than 1 min)</p> <ol style="list-style-type: none"> <li>a. Read the focus questions on the slide.</li> <li>a. <b>Pairs:</b> Have participants pair up with an elbow partner and share their initial ideas for answering the first focus question.</li> <li>b. <b>Whole group:</b> Invite participants to share their ideas with the group. During this share-out, record key ideas on chart paper.</li> </ol> <hr/> <p><b>Display Slide 27.</b> Investigating Physical Changes in Matter (25 min)</p> <p><b>Note:</b> Make sure to turn the griddle on, set it at low to medium heat, and let it warm up before the demonstration. Don't place the beakers containing the chocolate, crayons, and butter on the griddle until the demonstration begins. Caution participants not to get too close to the heated griddle or beakers.</p> <ol style="list-style-type: none"> <li>a. "In lessons 1 and 2 of this unit on matter, our students will explore physical changes that happen when heat is added to different kinds of matter. Today we'll explore some of these changes as well. Make sure to keep our first two-part focus question in mind throughout this investigation: <i>Are physical changes reversible at room temperature? What is your evidence?</i>"</li> <li>b. Have participants gather around the table for the demonstration. Place the three beakers containing the chocolate, crayons, and butter on the preheated griddle. The griddle should be warm enough to begin melting the substances. Ask participants to observe the substances</li> </ol>


PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
	<ul style="list-style-type: none"> <li>• RESPeCT lesson plans binder</li> </ul> <p><b>Resources in Lesson Plans Binder</b></p> <p><i>Resources section:</i></p> <ul style="list-style-type: none"> <li>• Content background document</li> <li>• Common Student Ideas</li> </ul>	<div data-bbox="848 695 1310 1013" style="border: 1px solid black; padding: 10px;"> <p style="text-align: center;"><b>What Would Your Students Say?</b></p> <p>How would your students describe the characteristics of solid butter and liquid butter? What science words might they use?</p>  <p>Think about this for a moment. Then write a possible student response in your science notebook.</p> </div>	<p>melting and record their observations in their science notebooks.</p> <p><b>Note:</b> If the chocolate and crayons aren't melting quickly enough, you may need to increase the temperature to 200 or 300 degrees.</p> <p>c. As each substance reaches its melting point, remove it from the griddle and place it on a pot holder to cool at room temperature. After removing all three substances from the heat, turn off the griddle.</p> <p>d. While the substances are cooling, advance to the next slide and discuss the question.</p> <hr/> <p><b>Display Slide 28.</b> What Would Your Students Say? (10 min)</p> <p>a. "As we give these substances time to cool, let's consider the questions on the slide: <i>How would your students describe the characteristics of solid butter and liquid butter? What science words might they use?</i> Reflect on this for a moment; then write a possible student response in your science notebooks."</p> <p>b. <b>Whole-group discussion:</b> Invite participants to share their ideas with the group. Feel free to contribute ideas based on your own classroom experiences. How do you think your own students would describe these substances? What science vocabulary do you think they'd use?</p> <p>c. After the group discussion, ask participants to examine the three substances that are cooling and share their observations. At this point, the crayons will likely be changing back to a solid, but the chocolate and butter may still be liquids.</p>

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
		<hr/> <p><b>NGSS Disciplinary Core Idea</b></p> <p><i>Different properties [of matter] are suited to different purposes.</i></p> <ul style="list-style-type: none"> <li>• How might solid butter and liquid butter be suited to different purposes?</li> <li>• At what temperature range do you think liquid butter would change back to solid butter?</li> <li>• How might this temperature range differ from the temperature necessary for liquid water or melted chocolate to change back to a solid?</li> </ul> 	<p><b>Display Slide 29.</b> NGSS Disciplinary Core Idea (7 min)</p> <p>a. Read the NGSS disciplinary core idea on the slide and ask participants the following questions:</p> <ul style="list-style-type: none"> <li>• How might solid butter and liquid butter be suited to different purposes?</li> <li>• At what temperature range do you think liquid butter would change back to solid butter?</li> <li>• How might this temperature range differ from the temperature necessary for liquid water or melted chocolate to change back to a solid?</li> </ul> <p>b. Elicit a variety of ideas from participants.</p>
		<hr/> <p><b>Unit Central Questions</b></p> <p>What is matter made of? How can matter change?</p> 	<p><b>Display Slide 30.</b> Unit Central Questions (Less than 1 min)</p> <p>a. Review the unit central questions on the slide.</p> <p>b. “Let’s see if we can gather some additional data that will help us develop a more complete answer to our second unit central question.”</p> <p>c. “But first, let’s review what we’ve learned so far about changes in matter.”</p>

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
		<p data-bbox="877 302 1266 326"><b>Review: Physical and Chemical Changes</b></p> <ul data-bbox="877 337 1255 380" style="list-style-type: none"> <li>• What is a physical change? What happens to the molecules in a physical change?</li> </ul>    <ul data-bbox="877 467 1255 509" style="list-style-type: none"> <li>• What is a chemical change? What happens to the molecules during a chemical reaction?</li> </ul> 	<p data-bbox="1335 272 1944 337"><b>Display Slide 31.</b> Review: Physical and Chemical Changes (10 min)</p> <ol data-bbox="1335 391 1944 724" style="list-style-type: none"> <li>“In our previous content deepening sessions, we investigated two types of changes in matter: physical changes and chemical changes.”</li> <li>“Who can describe what a physical change is? What happens to the molecules of a substance in a physical change? Does the substance itself change into something different? ”</li> <li>During this discussion, use probe and challenge questions to make participants’ thinking visible and ensure scientific accuracy.</li> </ol> <p data-bbox="1362 743 1560 768"><b>Ideal response:</b></p> <ul data-bbox="1383 776 1944 1320" style="list-style-type: none"> <li>• When heat is added to solid matter like ice, the molecules speed up, break away from their rigid structure, and begin moving around more freely as a liquid. In a liquid state, water molecules are still attracted to one another, but they can flow or slide past each other. Conversely, when heat is removed from liquid matter, the molecules slow down and join together in a rigid structure, where they vibrate in place as a solid. These physical changes in matter are reversible, so substances can change back and forth from solids to liquids and from liquids to solids when heat is added or removed. In physical changes the substance itself doesn’t change into a different substance; only the arrangement and movement of the molecules change.</li> </ul> <ol data-bbox="1335 1341 1944 1425" style="list-style-type: none"> <li>“Who can describe what a chemical change is? What happens to the atoms and molecules of a substance in a chemical change? Does the</li> </ol>

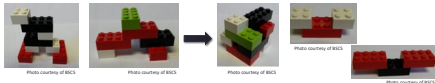
PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
			<p>substance itself remain the same or change into something different? “</p> <p>e. During this discussion, use probe and challenge questions to make participants’ thinking visible and ensure scientific accuracy.</p> <p><b>Ideal response:</b></p> <ul style="list-style-type: none"> <li>Substances like vinegar and baking soda can undergo a chemical change when they’re mixed together. The atoms of each molecule separate and recombine (rearrange) in different ways to form entirely new molecules or substances (types of matter) with different properties than the original substances. One of the substances produced in a chemical reaction between vinegar and baking soda is carbon dioxide (CO<sub>2</sub>). This invisible gas is proof that a chemical change has occurred.</li> </ul>
		<p><b>Reflect: Content Deepening Focus Question 1</b></p> <p>Are physical changes reversible at room temperature? What is your evidence?</p>	<p><b>Display Slide 32.</b> Reflect: Content Deepening Focus Question 1 (7 min)</p> <p>a. Review the focus questions on the slide.</p> <p>b. <b>Individuals:</b> Ask participants to answer these questions in their science notebooks, using evidence from the demonstration of melting and freezing and science ideas from previous content deepening sessions.</p> <p>c. <b>Whole group:</b> Invite a few participants to share their answers to the focus questions. Record key ideas on chart paper.</p>
<b>10-MINUTE BREAK</b>			

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
		<p style="text-align: center;"><b>Content Deepening: Focus Question 2</b></p> <p>Are atoms created or destroyed when matter undergoes a physical or chemical change? How can you prove your ideas?</p>	<p><b>Display Slide 33.</b> Content Deepening: Focus Question 2 (7 min)</p> <ol style="list-style-type: none"> <li>Read the focus questions on the slide.</li> <li>“This focus question requires us to prove something about molecules even though we can’t see them. But the beauty is that this is possible!”</li> <li><b>Pairs:</b> Have participants pair up with an elbow partner and discuss their ideas for answering this question and ways they could prove their ideas are right. Encourage participants to consider examples of physical and chemical changes from content deepening work over the past few days.</li> <li><b>Whole group:</b> Invite participants to share their initial ideas for answering the focus question and ways they could prove their ideas are right. Record participants’ ideas on chart paper.</li> </ol>
		<p style="text-align: center;"><b>Design an Experiment</b></p> <div style="text-align: center;">  <p>Physical Change</p> <p>→ Add Heat →</p> <p>← Remove Heat ←</p> </div> <div style="text-align: center;">  <p>Chemical Change</p> </div>	<p><b>Display Slide 34.</b> Let’s Design an Experiment! (10 min)</p> <ol style="list-style-type: none"> <li>“Since we can’t see molecules, we’ll use our Lego model again to help us design an experiment that will help us prove our ideas about whether atoms are created or destroyed when matter undergoes a physical or chemical change.”</li> <li>Using the Lego images on the slide, work together as a group to design an experiment that will accomplish the specified goal.</li> </ol> <p><b>Note:</b> Accept all ideas at this point. The next few slides will help you acquaint participants with a specific experimental method.</p> <ol style="list-style-type: none"> <li>During this design process, you may need to</li> </ol>

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
		<p data-bbox="871 800 1222 824"><b>Testing Our Ideas: Physical Changes</b></p> <p data-bbox="871 846 1268 915">Are atoms created or destroyed when matter undergoes a physical or chemical change? How can you prove your ideas?</p> 	<p data-bbox="1362 256 1848 313">probe participants' understandings of the following key science ideas:</p> <ul data-bbox="1383 334 1948 724" style="list-style-type: none"> <li>• In a physical change, the molecules aren't rearranged. They simply occupy more or less space based on their arrangement and movement.</li> <li>• In a chemical change, a reaction occurs in which the atoms of the original substances separate and recombine or rearrange to form new molecules or substances with different properties.</li> <li>• In both types of change, the number of atoms before the change equals the number of atoms after the change. This can be measured!</li> </ul> <p data-bbox="1335 764 1894 824"><b>Display Slide 35.</b> Testing Our Ideas: Physical Changes (Less than 1 min)</p> <p data-bbox="1335 881 1940 1060">a. "First, let's focus on physical changes in matter. Using a balance, we'll test our ideas or hypotheses and make observations to determine with confidence whether atoms are created or destroyed or remain the same when matter undergoes a physical change."</p>

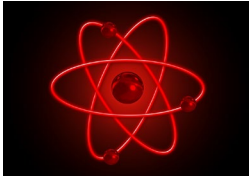


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		<p style="text-align: center;"><b>Testing Our Ideas: Physical Changes</b></p> <ul style="list-style-type: none"> <li>• <b>Both teams:</b> Set up your balance scales. Then using Team 1's Legos, build a total of 12 Mickey Mouse water molecules.</li> <li>• <b>Team 1:</b> Place 6 Lego water molecules on one side of your balance and 6 molecules on the other side.</li> <li>• <b>Team 2:</b> Wipe any condensation off your frozen water bottle. Then place the frozen bottle on one side of your balance and the bottle at room temperature on the other side.</li> <li>• <b>Both teams:</b> Observe and interpret the weight results.</li> </ul>	<p><b>Display Slide 36.</b> Testing Our Ideas: Physical Changes (15 min)</p> <p>a. Divide the group into two teams. Give the first team a balance scale, 24 white (hydrogen) Legos, and 12 red (oxygen) Legos. Give the second team a balance scale, one bottle of frozen water, and one bottle of room-temperature water.</p> <p>b. Direct teams to follow the instructions on the slide.</p> <ul style="list-style-type: none"> <li>• Both teams should set up their balance scales and then Team 2 should help Team 1 build a total of 12 Mickey Mouse Lego water molecules (six each).</li> <li>• Team 1 should weigh their Lego water molecules by placing six molecules on one side of their balance and six molecules on the other side. Team 2 should weigh their water bottles by placing the frozen bottle on one side of their balance and the room-temperature bottle on the other side. It may be easier if they lay the bottles on their sides instead of placing them upright. <b>Note: Make sure participants wipe any condensation off the outside of the frozen bottle to avoid skewing the results.</b></li> <li>• Both teams should observe, record, and interpret the results.</li> </ul> <p>c. <b>Whole group:</b> Ask participants how these experiments relate to the focus question, <i>Are atoms created or destroyed when matter undergoes a physical or chemical change?</i> Did the results prove their ideas or hypotheses?</p> <p><b>Ideal response:</b></p> <ul style="list-style-type: none"> <li>• Since both sides of each balance are equal or</li> </ul>

PD Model: Time/Phase	Purpose, Content, and What Participants Do	Slides	Process
		<p data-bbox="877 467 1241 493"><b>Testing Our Ideas: Chemical Changes</b></p> <ul data-bbox="877 505 1289 678" style="list-style-type: none"> <li>• <b>Team 1:</b> Build Lego models of the two <b>reagent</b> molecules (vinegar, baking soda). Place your molecules on one side of the balance scale.</li> <li>• <b>Team 2:</b> Build Lego models of the three <b>product</b> molecules (sodium acetate, water, carbon dioxide). Place your molecules on the other side of the balance scale.</li> </ul> 	<p data-bbox="1360 253 1944 402">level, this proves that atoms aren't created or destroyed when water undergoes a physical change. The molecules of solid water are exactly the same weight as the molecules of liquid water. Matter is conserved, so the scales balanced.</p> <p data-bbox="1333 440 1906 500"><b>Display Slide 37.</b> Testing Our Ideas: Chemical Changes (7 min)</p> <p data-bbox="1333 558 1934 675">a. "Next, we'll focus on chemical changes in matter and use a balance to determine whether atoms are created or destroyed or remain the same when matter undergoes a chemical change."</p> <p data-bbox="1333 695 1906 721">b. Walk teams through the following instructions:</p> <ul data-bbox="1381 727 1955 1419" style="list-style-type: none"> <li>• Set up one balance scale.</li> <li>• Have Team 1 build Lego models of the two reagent molecules shown on the slide. <ul data-bbox="1430 818 1934 1029" style="list-style-type: none"> <li>• <i>Vinegar molecule:</i> four white (hydrogen) Legos, two red (oxygen) Legos, and two black (carbon) Legos</li> <li>• <i>Baking-soda molecule:</i> one lime-green (sodium) Lego, one white (hydrogen) Lego, three red (oxygen) Legos, and one black (carbon) Lego</li> </ul> </li> <li>• Have Team 2 set up their balance scale and build Lego models of the three products of the neutralization reaction shown on the slide. <ul data-bbox="1430 1127 1934 1386" style="list-style-type: none"> <li>• <i>Sodium-acetate molecule:</i> one lime-green (sodium) Lego, three white (hydrogen) Legos, two red (oxygen) Legos, and two black (carbon) Legos</li> <li>• <i>Water molecule:</i> two white (hydrogen) Legos and one red (oxygen) Lego</li> <li>• <i>Carbon-dioxide molecule:</i> one black (carbon) Lego and two red (oxygen) Legos</li> </ul> </li> <li>• Direct Team 1 to place their two reagent Lego</li> </ul>

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		<p style="text-align: center;"><b>Testing Our Ideas: Chemical Changes</b></p> <ul style="list-style-type: none"> <li>• <b>Both teams:</b> Place 1 teaspoon of baking soda in each freezer bag. Place 2 tablespoons of vinegar in each vial and secure the cap. Wipe off any vinegar on the outside of the vials. Then place the vials in the freezer bags and seal them.</li> <li>• <b>Team 1:</b> Place one of the sealed freezer bags on one side of the balance. <b>Don't open the bag or take the cap off the vial.</b></li> <li>• <b>Team 2:</b> <b>Without opening the second bag,</b> pop the cap off the vial and mix the vinegar with the baking soda. When the bag puffs up with gas, place the bag on the other side of the balance.</li> </ul>	<p>molecules on one side of the balance scale, and have Team 2 place their three product molecules on the other side.</p> <ul style="list-style-type: none"> <li>• Ask the teams to observe, record, and interpret the results.</li> </ul> <p><b>Display Slide 38.</b> Testing Our Ideas: Chemical Changes (10 min)</p> <p>a. "Next, we'll weigh the actual substances in this chemical change and see if we get the same results."</p> <p>b. Walk teams through the following instructions for preparing identical freezer bags containing baking soda and vinegar:</p> <ul style="list-style-type: none"> <li>• Place 1 teaspoon of baking soda in each bag.</li> <li>• Place 2 tablespoons of vinegar in each plastic vial. Then secure the lids and wipe any vinegar off the outside of each vial.</li> <li>• Place the vials in the freezer bags and remove as much air from the bags before sealing them.</li> <li>• Have Team 1 place one of the freezer bags on one side of the balance scale <i>without mixing the substances together</i>. <b>Caution participants not to pop off the cap on the vial of vinegar.</b></li> <li>• For the second freezer bag, have Team 2 pop the cap off the vial of vinegar and mix the substances together <i>without opening the bag</i>. (Make sure participants understand that the bag should remain sealed.) After the chemical reaction occurs and the bag puffs up with gas, have Team 2 place the bag on the other side of the balance scale.</li> <li>• Ask the teams to observe, record, and interpret the results.</li> </ul>

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			<p>c. <b>Whole group:</b> Ask participants how this experiment relates to the focus question, <i>Are atoms created or destroyed when matter undergoes a physical or chemical change?</i> Did the results prove their ideas or hypotheses?</p> <p><b>Ideal response:</b></p> <ul style="list-style-type: none"> <li>• Since both sides of the scale are balanced (equal), this proves that atoms aren't created or destroyed when a chemical change occurs. The reagent molecules (before the reaction) are exactly the same weight as the product molecules (after the reaction). Matter is conserved, so the scale balanced.</li> </ul>
		<hr style="border: 2px solid #808080;"/> <p><b>Reflect: Content Deepening Focus Question 2</b></p> <p>Are atoms created or destroyed when matter undergoes a physical or chemical change? How can you prove your ideas?</p>	<p><b>Display Slide 39.</b> Reflect: Content Deepening Focus Question 2 (7 min)</p> <p>a. Review the two-part focus question on the slide.</p> <p>b. <b>Individuals:</b> Ask participants to answer these questions in their science notebooks, using the data they just collected from the experiments on physical and chemical changes.</p> <p>c. <b>Whole group:</b> Invite a few participants to share their answers to the focus questions. Record key ideas on chart paper.</p>

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		<p><b>NGSS Connections</b></p> <p>Disciplinary core ideas (2-PS1-1; 2-PS1-4):</p> <ol style="list-style-type: none"> <li>1. Different kinds of matter exist and many of them can be either solid or liquid, depending on temperature. Matter can be described and classified by its observable properties.</li> <li>2. Heating or cooling a substance may cause changes that can be observed. Sometimes these changes are reversible, and sometimes they are not.</li> </ol> <ul style="list-style-type: none"> <li>• How did today's content deepening activities address these core ideas?</li> <li>• How would addressing these ideas in the classroom be valuable for our students?</li> </ul>	<p><b>Display Slide 40.</b> NGSS Connections (10 min)</p> <ol style="list-style-type: none"> <li>a. Review the NGSS disciplinary core ideas on the slide. Then ask participants, "How did today's content deepening activities address these core ideas? How would addressing these ideas in the classroom be valuable for our students?"</li> <li>b. <b>Individuals:</b> Ask participants to think about these questions and then write their ideas in their science notebooks.</li> <li>c. <b>Whole group:</b> Invite participants to share their responses with the group. Record key ideas on chart paper.</li> </ol>
		<p><b>Unit Central Questions</b></p> <p>What is matter made of? How can matter change?</p> 	<p><b>Display Slide 41.</b> Unit Central Questions (5 min)</p> <ol style="list-style-type: none"> <li>a. Review the unit central questions on the slide.</li> <li>b. Ask participants, "How would you answer these questions based on today's content deepening work? What new ideas can you add?"</li> <li>c. Avoid adding on to participants' responses. Simply ask probe and challenge questions to ensure that participants express their ideas about matter in scientifically accurate ways.</li> </ol> <p><b>Note:</b> Ideally, participants will mention the arrangement and movement of molecules in physical changes and the rearrangement of atoms and molecules in chemical changes without losing or gaining any atoms (conservation of matter). If needed, ask participants the challenge question, "Can the changes in matter be qualified?"</p>

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<p>3:15–3:30 15 min</p> <p><b>Wrap-Up: Summary, Homework, and Reflections</b></p> <p>Slides 42–45</p>	<p><b>Purpose</b></p> <ul style="list-style-type: none"> <li>• Reflect on the day’s learning and summarize key ideas about the science content and strategies 4 and 5, linking those ideas to participants’ images of effective science teaching and changes they want to make in their individual teaching practices.</li> </ul> <p><b>What Participants Do</b></p> <ul style="list-style-type: none"> <li>• Discuss ways of moving student thinking forward.</li> <li>• Add to/modify the Effective Science Teaching chart.</li> <li>• Review and discuss (as needed) today’s focus questions.</li> <li>• Learn about the homework assignment and the focus of tomorrow’s work.</li> <li>• Write reflections on today’s learning.</li> </ul> <p><b>Posters and Charts</b></p> <ul style="list-style-type: none"> <li>• Effective Science Teaching chart</li> <li>• Strategy charts created today for STL strategies 4 and 5</li> </ul> <p><b>Handouts in PD Binder</b></p> <ul style="list-style-type: none"> <li>• 3.5 Daily Reflections—Day 3</li> </ul> <p><b>Supplies</b></p> <ul style="list-style-type: none"> <li>• Science notebooks</li> </ul> <p><b>PD Resources</b></p> <ul style="list-style-type: none"> <li>• STeLLA strategies booklet</li> </ul>	<p><b>Summary: Moving Student Thinking Forward</b></p> <ol style="list-style-type: none"> <li>1. How can we advance student thinking without simply telling students about science ideas and asking them to memorize the concepts?</li> <li>2. Refer to our Effective Science Teaching chart from day 1. Which of these ideas do you want to highlight based on the strategies we’ve explored so far? Anything you want to add or modify?</li> </ol>	<p><b>Display Slide 42.</b> Summary: Moving Student Thinking Forward (5 min)</p> <ol style="list-style-type: none"> <li>a. Have participants share ideas about the first question on the slide. Then ask, “What are some things we’ve discussed today that address this question?”</li> <li>b. Refer participants to the Effective Science Teaching chart from day 1 and discuss the remaining questions on the slide. Modify the chart as participants share their ideas.</li> </ol>		
		<p><b>Today’s Focus Questions</b></p> <table border="0"> <tr> <td data-bbox="871 844 1071 1031"> <p><b>Lesson Analysis</b></p> <ul style="list-style-type: none"> <li>• How can analyzing data and constructing explanations help students <b>move forward</b> toward deeper understandings of science ideas?</li> </ul> </td> <td data-bbox="1081 844 1291 1096"> <p><b>Content Deepening</b></p> <ul style="list-style-type: none"> <li>• Are physical changes reversible at room temperature? What is your evidence?</li> <li>• Are atoms created or destroyed when matter undergoes a physical or chemical change? How can you prove your ideas?</li> </ul> </td> </tr> </table>	<p><b>Lesson Analysis</b></p> <ul style="list-style-type: none"> <li>• How can analyzing data and constructing explanations help students <b>move forward</b> toward deeper understandings of science ideas?</li> </ul>	<p><b>Content Deepening</b></p> <ul style="list-style-type: none"> <li>• Are physical changes reversible at room temperature? What is your evidence?</li> <li>• Are atoms created or destroyed when matter undergoes a physical or chemical change? How can you prove your ideas?</li> </ul>	<p><b>Display Slide 43.</b> Today’s Focus Questions (5 min)</p> <ol style="list-style-type: none"> <li>a. Review today’s focus questions.</li> <li>b. <b>Discuss:</b> “The STeLLA strategies booklet claims that strategies 4 and 5 are ways of moving student thinking forward. How would you support or challenge that claim? In other words, are you convinced that letting students analyze data and construct explanations will help them move forward toward deeper understandings of science ideas?”</li> <li>c. <b>Ask:</b> “What key ideas do you now have about how to address our content deepening focus questions?”</li> </ol>
<p><b>Lesson Analysis</b></p> <ul style="list-style-type: none"> <li>• How can analyzing data and constructing explanations help students <b>move forward</b> toward deeper understandings of science ideas?</li> </ul>	<p><b>Content Deepening</b></p> <ul style="list-style-type: none"> <li>• Are physical changes reversible at room temperature? What is your evidence?</li> <li>• Are atoms created or destroyed when matter undergoes a physical or chemical change? How can you prove your ideas?</li> </ul>				

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	<ul style="list-style-type: none"> <li>STL Z-fold summary chart (front pocket of PD binder)</li> </ul>	<hr/> <p><b>Homework</b></p> <ol style="list-style-type: none"> <li>Review strategy 6 in the STeLLA strategies booklet and complete the STL Z-fold summary chart for this strategy; Engage students in using and applying new science ideas in a variety of ways and contexts.</li> <li>Be prepared to share your assigned lesson plan review.</li> </ol>	<p><b>Display Slide 44.</b> Homework (2 min)</p> <ol style="list-style-type: none"> <li>“Tomorrow we’ll focus on another strategy to help move student thinking forward toward deeper understandings of science ideas.”</li> <li>Review the homework assignment and have participants copy it into their science notebooks.</li> </ol>
		<hr/> <p><b>Reflections on Today’s Session</b></p> <p>Complete the Daily Reflections sheet (handout 3.5).</p> <ol style="list-style-type: none"> <li>What new idea or insight did you have today related to strategy 4 (analyzing and interpreting data and observations) and strategy 5 (constructing explanations and arguments)?</li> <li>What ideas do strategies 4 and 5 give you about things to try or change in your science teaching?</li> <li>Answer one of these questions: (1) What important science idea are you taking away from our content deepening work today? Remember to state the idea in a complete sentence. (2) What question do you have about matter and how it can change (i.e., something you’re unclear or wonder about)?</li> </ol>	<p><b>Display Slide 45.</b> Reflection on Today’s Session (3 min)</p> <ol style="list-style-type: none"> <li>Have participants reflect on today’s session and answer the questions on the Daily Reflections sheet (handout 3.5 in PD program binder).</li> </ol> <p><b>Note:</b> To support this task, encourage participants to refer to the STeLLA strategies booklet, the charts they created for STL strategies 4 and 5, the Effective Science Teaching chart, and their STL Z-fold summary charts.</p>